

# Glutaric acid, 2,5-dichlorophenyl propyl ester

<b>Inchi:</b>	InChI=1S/C14H16Cl2O4/c1-2-8-19-13(17)4-3-5-14(18)20-12-9-10(15)6-7-11(12)16/h6-7,
<b>InchiKey:</b>	XMOJJPSPLYDERT-UHFFFAOYSA-N
<b>Formula:</b>	C14H16Cl2O4
<b>SMILES:</b>	CCCOC(=O)CCCC(=O)Oc1cc(Cl)ccc1Cl
<b>Mol. weight [g/mol]:</b>	319.18

## Physical Properties

Property code	Value	Unit	Source
gf	-331.55	kJ/mol	Joback Method
hf	-639.78	kJ/mol	Joback Method
hfus	39.25	kJ/mol	Joback Method
hvap	77.44	kJ/mol	Joback Method
log10ws	-4.53		Crippen Method
logp	4.022		Crippen Method
mcvol	223.720	ml/mol	McGowan Method
pc	1987.66	kPa	Joback Method
rinpola	2212.00		NIST Webbook
tb	783.80	K	Joback Method
tc	997.33	K	Joback Method
tf	503.16	K	Joback Method
vc	0.858	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	588.93	J/molxK	783.80	Joback Method
cpg	601.07	J/molxK	819.39	Joback Method
cpg	612.28	J/molxK	854.98	Joback Method
cpg	622.58	J/molxK	890.56	Joback Method
cpg	631.97	J/molxK	926.15	Joback Method
cpg	640.45	J/molxK	961.74	Joback Method
cpg	648.04	J/molxK	997.33	Joback Method
dvisc	0.0006262	Paxs	503.16	Joback Method
dvisc	0.0004009	Paxs	549.93	Joback Method

dvisc	0.0002752	Paxs	596.71	Joback Method
dvisc	0.0001996	Paxs	643.48	Joback Method
dvisc	0.0001511	Paxs	690.25	Joback Method
dvisc	0.0001186	Paxs	737.03	Joback Method
dvisc	0.0000958	Paxs	783.80	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U358993&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U358993&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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