

# 1-Cyclohexene, 3-methoxy-

<b>Other names:</b>	Cyclohexene, 3-methoxy-
<b>Inchi:</b>	InChI=1S/C7H12O/c1-8-7-5-3-2-4-6-7/h3,5,7H,2,4,6H2,1H3
<b>InchiKey:</b>	OKDKFTKUXADLSJ-UHFFFAOYSA-N
<b>Formula:</b>	C7H12O
<b>SMILES:</b>	COC1C=CCCC1
<b>Mol. weight [g/mol]:</b>	112.17
<b>CAS:</b>	2699-13-0

## Physical Properties

Property code	Value	Unit	Source
gf	-42.53	kJ/mol	Joback Method
hf	-207.93	kJ/mol	Joback Method
hfus	8.13	kJ/mol	Joback Method
hvap	34.31	kJ/mol	Joback Method
log10ws	-1.70		Crippen Method
logp	1.742		Crippen Method
mcvol	100.200	ml/mol	McGowan Method
pc	3602.88	kPa	Joback Method
tb	400.69	K	Joback Method
tc	604.95	K	Joback Method
tf	199.02	K	Joback Method
vc	0.364	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	184.89	J/mol×K	400.69	Joback Method
cpg	251.04	J/mol×K	570.90	Joback Method
cpg	239.05	J/mol×K	536.86	Joback Method
cpg	226.45	J/mol×K	502.82	Joback Method
cpg	213.23	J/mol×K	468.78	Joback Method
cpg	199.38	J/mol×K	434.73	Joback Method
cpg	262.43	J/mol×K	604.95	Joback Method
dvisc	0.0002392	Paxs	400.69	Joback Method

dvisc	0.0003089	Paxs	367.08	Joback Method
dvisc	0.0004202	Paxs	333.47	Joback Method
dvisc	0.0006123	Paxs	299.86	Joback Method
dvisc	0.0009812	Paxs	266.24	Joback Method
dvisc	0.0018021	Paxs	232.63	Joback Method
dvisc	0.0040640	Paxs	199.02	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2699130&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2699130&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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