

# Formic acid, 3-phenylpropyl ester

<b>Inchi:</b>	InChI=1S/C10H12O2/c11-9-12-8-4-7-10-5-2-1-3-6-10/h1-3,5-6,9H,4,7-8H2
<b>InchiKey:</b>	AMHRXSOVSAQOKL-UHFFFAOYSA-N
<b>Formula:</b>	C10H12O2
<b>SMILES:</b>	O=COCCc1ccccc1
<b>Mol. weight [g/mol]:</b>	164.20

## Physical Properties

Property code	Value	Unit	Source
gf	-58.79	kJ/mol	Joback Method
hf	-231.00	kJ/mol	Joback Method
hfus	19.17	kJ/mol	Joback Method
hvap	49.26	kJ/mol	Joback Method
log10ws	-1.97		Crippen Method
logp	1.792		Crippen Method
mvol	135.440	ml/mol	McGowan Method
pc	3121.00	kPa	Joback Method
rinpol	1309.00		NIST Webbook
rinpol	1309.00		NIST Webbook
tb	525.96	K	Joback Method
tc	733.48	K	Joback Method
tf	293.11	K	Joback Method
vc	0.522	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	301.53	J/molxK	525.96	Joback Method
cpg	360.65	J/molxK	698.89	Joback Method
cpg	350.21	J/molxK	664.31	Joback Method
cpg	339.09	J/molxK	629.72	Joback Method
cpg	327.29	J/molxK	595.13	Joback Method
cpg	314.77	J/molxK	560.55	Joback Method
cpg	370.44	J/molxK	733.48	Joback Method
dvisc	0.0002358	Paxs	525.96	Joback Method

dvisc	0.0003005	Paxs	487.15	Joback Method
dvisc	0.0003993	Paxs	448.34	Joback Method
dvisc	0.0005600	Paxs	409.54	Joback Method
dvisc	0.0008430	Paxs	370.73	Joback Method
dvisc	0.0013964	Paxs	331.92	Joback Method
dvisc	0.0026437	Paxs	293.11	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U367887&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U367887&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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