

# 2-Propanone, methylhydrazone

<b>Other names:</b>	Acetone methylhydrazone
<b>Inchi:</b>	InChI=1S/C4H10N2/c1-4(2)6-5-3/h5H,1-3H3
<b>InchiKey:</b>	JXSSEZYJBTUIGX-UHFFFAOYSA-N
<b>Formula:</b>	C4H10N2
<b>SMILES:</b>	CNN=C(C)C
<b>Mol. weight [g/mol]:</b>	86.14
<b>CAS:</b>	5771-02-8

## Physical Properties

Property code	Value	Unit	Source
chl	-3043.90 ± 0.80	kJ/mol	NIST Webbook
hf	0.01	kJ/mol	Joback Method
hfl	39.70 ± 0.80	kJ/mol	NIST Webbook
hvap	34.33	kJ/mol	Joback Method
ie	7.69	eV	NIST Webbook
log10ws	-0.84		Crippen Method
logp	0.602		Crippen Method
mcvol	82.880	ml/mol	McGowan Method
pc	3505.43	kPa	Joback Method
rinpol	728.00		NIST Webbook
rinpol	728.00		NIST Webbook
tb	417.65	K	Joback Method
tc	616.64	K	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5771028&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5771028&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

# Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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