

# Benzeneethanol, 4-methyl-

<b>Other names:</b>	Phenethyl alcohol, p-methyl- p-Methylphenethyl alcohol 2-(p-Methylphenyl)ethanol 2-(4-Methylphenyl)ethanol 4-Methylbenzeneethanol 4-Methylphenethyl alcohol p-Methylphenylethyl alcohol 2-p-Tolyloethanol NSC 5295 NSC 54360
<b>Inchi:</b>	InChI=1S/C9H12O/c1-8-2-4-9(5-3-8)6-7-10/h2-5,10H,6-7H2,1H3
<b>InchiKey:</b>	DAVFJRVIVZOKKS-UHFFFAOYSA-N
<b>Formula:</b>	C9H12O
<b>SMILES:</b>	<chem>Cc1ccc(CCO)cc1</chem>
<b>Mol. weight [g/mol]:</b>	136.19
<b>CAS:</b>	699-02-5

## Physical Properties

Property code	Value	Unit	Source
gf	-9.14	kJ/mol	Joback Method
hf	-156.26	kJ/mol	Joback Method
hfus	16.81	kJ/mol	Joback Method
hvap	55.25	kJ/mol	Joback Method
log10ws	-2.01		Crippen Method
logp	1.530		Crippen Method
mcvol	119.780	ml/mol	McGowan Method
pc	3568.53	kPa	Joback Method
tb	517.70	K	NIST Webbook
tc	725.15	K	Joback Method
tf	290.95	K	Joback Method
vc	0.451	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	266.45	J/mol×K	529.16	Joback Method
cpg	277.78	J/mol×K	561.83	Joback Method
cpg	288.52	J/mol×K	594.49	Joback Method
cpg	298.68	J/mol×K	627.16	Joback Method
cpg	308.29	J/mol×K	659.82	Joback Method
cpg	317.36	J/mol×K	692.49	Joback Method
cpg	325.92	J/mol×K	725.15	Joback Method
dvisc	0.0098792	Paxs	290.95	Joback Method
dvisc	0.0030353	Paxs	330.65	Joback Method
dvisc	0.0012010	Paxs	370.35	Joback Method
dvisc	0.0005687	Paxs	410.05	Joback Method
dvisc	0.0003073	Paxs	449.76	Joback Method
dvisc	0.0001835	Paxs	489.46	Joback Method
dvisc	0.0001184	Paxs	529.16	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C699025&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C699025&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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