

P-TOLYLMETHANETHIOL

Other names:	p-Xylene-alpha-thiol p-xylene-«alpha»-thiol
Inchi:	InChI=1S/C8H10S/c1-7-2-4-8(6-9)5-3-7/h2-5,9H,6H2,1H3
InchiKey:	AGFYZLVFPSGUIX-UHFFFAOYSA-N
Formula:	C8H10S
SMILES:	Cc1ccc(CS)cc1
Mol. weight [g/mol]:	138.24

Physical Properties

Property code	Value	Unit	Source
af	0.4268		Cheméo Relay (1.0)
gf	148.65	kJ/mol	Joback Method
hf	41.13	kJ/mol	Cheméo Relay (1.0)
hfus	14.17	kJ/mol	Joback Method
hvap	60.70	kJ/mol	Cheméo Relay (1.0)
ie	8.34	eV	Cheméo Relay (1.0)
log10ws	-2.45		Cheméo Relay (1.0)
logp	2.425		Crippen Method
mcvol	116.170	ml/mol	McGowan Method
pc	3877.12	kPa	Joback Method
tb	481.82	K	Cheméo Relay (1.0)
tc	710.45	K	Cheméo Relay (1.0)
tf	261.65	K	Cheméo Relay (1.0)
vc	0.418	m3/kmol	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	221.48	J/mol×K	476.96	Joback Method
cpg	234.25	J/mol×K	516.37	Joback Method
cpg	246.23	J/mol×K	555.77	Joback Method
cpg	257.43	J/mol×K	595.18	Joback Method
cpg	267.90	J/mol×K	634.58	Joback Method
cpg	277.66	J/mol×K	673.99	Joback Method

Sources

Cheméo Relay (1.0, beta):

NIST Webbook: <http://webbook.nist.gov/cgi/cbook.cgi?ID=C4498991&Units=SI>

Joback Method: https://en.wikipedia.org/wiki/Joback_method

McGowan Method: <http://link.springer.com/article/10.1007/BF02311772>

Crippen Method: <http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method: https://www.chemeo.com/doc/models/crippen_log10ws

Cheméo Relay (1.0):

Legend

af:	Acentric Factor
cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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