

# 3,4-Dimethylbenzyl isothiocyanate

Inchi:	InChI=1S/C10H11NS/c1-8-3-4-10(5-9(8)2)6-11-7-12/h3-5H,6H2,1-2H3
InchiKey:	BNOGRVSYSGZCDC-UHFFFAOYSA-N
Formula:	C10H11NS
SMILES:	Cc1ccc(CN=C=S)cc1C
Mol. weight [g/mol]:	177.27
CAS:	206559-59-3

## Physical Properties

Property code	Value	Unit	Source
hf	247.93	kJ/mol	Joback Method
hvap	51.89	kJ/mol	Joback Method
log10ws	-3.65		Crippen Method
logp	2.906		Crippen Method
mcvol	145.730	ml/mol	McGowan Method
pc	2906.11	kPa	Joback Method
tb	610.79	K	Joback Method
tc	861.13	K	Joback Method

## Sources

McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C206559593&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C206559593&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l

<b>logP:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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