

# exo-2-Hydroxycineole acetate

<b>Other names:</b>	1,3,3-Trimethyl-2-oxabicyclo[2.2.2]octan-6-ol, acetate 2-acetoxy-1,8-cineole exo-2-hydroxycineol acetate 1,3,3-trimethyl-2-oxabicyclo[2.2.2]octan-6-yl acetate 2-Hydroxycineole, acetate
<b>Inchi:</b>	InChI=1S/C12H20O3/c1-8(13)14-10-7-9-5-6-12(10,4)15-11(9,2)3/h9-10H,5-7H2,1-4H3
<b>InchiKey:</b>	XRKZFWIYZDOQO-UHFFFAOYSA-N
<b>Formula:</b>	C12H20O3
<b>SMILES:</b>	CC(=O)OC1CC2CCC1(C)OC2(C)C
<b>Mol. weight [g/mol]:</b>	212.29
<b>CAS:</b>	72257-53-5

## Physical Properties

Property code	Value	Unit	Source
gf	-198.98	kJ/mol	Joback Method
hf	-544.73	kJ/mol	Joback Method
hfus	19.22	kJ/mol	Joback Method
hvap	53.22	kJ/mol	Joback Method
log10ws	-2.68		Crippen Method
logp	2.286		Crippen Method
mcvol	171.530	ml/mol	McGowan Method
pc	2502.50	kPa	Joback Method
rinpol	1352.00		NIST Webbook
rinpol	1367.60		NIST Webbook
rinpol	1345.00		NIST Webbook
rinpol	1342.00		NIST Webbook
rinpol	1339.00		NIST Webbook
rinpol	1342.00		NIST Webbook
rinpol	1344.00		NIST Webbook
rinpol	1354.00		NIST Webbook
tb	590.36	K	Joback Method
tc	811.40	K	Joback Method
tf	391.89	K	Joback Method
vc	0.644	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	472.00	J/mol×K	590.36	Joback Method
cpg	490.68	J/mol×K	627.20	Joback Method
cpg	508.27	J/mol×K	664.04	Joback Method
cpg	525.00	J/mol×K	700.88	Joback Method
cpg	541.09	J/mol×K	737.72	Joback Method
cpg	556.75	J/mol×K	774.56	Joback Method
cpg	572.21	J/mol×K	811.40	Joback Method

## Sources

Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C72257535&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C72257535&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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