

5-ethyl-4-methyl-2-pentyl-3-thiazoline, cis

Inchi:	InChI=1S/C11H21NS/c1-4-6-7-8-11-12-9(3)10(5-2)13-11/h10-11H,4-8H2,1-3H3/t10-,11+
InchiKey:	WJKRIVHZAXUEOL-MNOVXSKEA-N
Formula:	C11H21NS
SMILES:	CCCCC1N=C(C)C(CC)S1
Mol. weight [g/mol]:	199.36

Physical Properties

Property code	Value	Unit	Source
gf	247.55	kJ/mol	Joback Method
hf	-67.69	kJ/mol	Joback Method
hfus	28.88	kJ/mol	Joback Method
hvap	53.00	kJ/mol	Joback Method
log10ws	-4.09		Crippen Method
logp	3.879		Crippen Method
mcvol	177.020	ml/mol	McGowan Method
pc	2244.00	kPa	Joback Method
rinpol	1485.00		NIST Webbook
rinpol	1481.00		NIST Webbook
rinpol	1485.00		NIST Webbook
ripol	1844.00		NIST Webbook
ripol	1844.00		NIST Webbook
tb	567.36	K	Joback Method
tc	778.45	K	Joback Method
tf	388.66	K	Joback Method
vc	0.672	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	451.92	J/molxK	567.36	Joback Method
cpg	471.24	J/molxK	602.54	Joback Method
cpg	489.55	J/molxK	637.72	Joback Method
cpg	506.88	J/molxK	672.90	Joback Method
cpg	523.23	J/molxK	708.09	Joback Method

cpg	538.63	J/mol×K	743.27	Joback Method
cpg	553.08	J/mol×K	778.45	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R498349&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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