

# Diethylmalonic acid, heptadecyl 3-methylbenzyl ester

Inchi:	InChI=1S/C32H54O4/c1-5-8-9-10-11-12-13-14-15-16-17-18-19-20-21-25-35-30(33)32(6-
InchiKey:	IHBIHZPYXHRFCO-UHFFFAOYSA-N
Formula:	C32H54O4
SMILES:	CCCCCCCCCCCCCCCCOC(=O)C(CC)(CC)C(=O)OCc1cccc(C)c1
Mol. weight [g/mol]:	502.77

## Physical Properties

Property code	Value	Unit	Source
gf	-143.66	kJ/mol	Joback Method
hf	-977.10	kJ/mol	Joback Method
hfus	70.45	kJ/mol	Joback Method
hvap	106.78	kJ/mol	Joback Method
log10ws	-10.36		Crippen Method
logp	9.259		Crippen Method
mcvol	452.860	ml/mol	McGowan Method
pc	670.12	kPa	Joback Method
rinpol	3375.00		NIST Webbook
tb	1112.57	K	Joback Method
tc	1382.84	K	Joback Method
tf	636.08	K	Joback Method
vc	1.756	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1625.49	J/molxK	1112.57	Joback Method
cpg	1707.47	J/molxK	1337.80	Joback Method
cpg	1694.46	J/molxK	1292.75	Joback Method
cpg	1679.92	J/molxK	1247.71	Joback Method
cpg	1663.70	J/molxK	1202.66	Joback Method
cpg	1645.62	J/molxK	1157.62	Joback Method
cpg	1719.14	J/molxK	1382.84	Joback Method
dvisc	0.0000065	Paxs	1112.57	Joback Method
dvisc	0.0000088	Paxs	1033.15	Joback Method

dvisc	0.0000124	Paxs	953.74	Joback Method
dvisc	0.0000187	Paxs	874.32	Joback Method
dvisc	0.0000304	Paxs	794.91	Joback Method
dvisc	0.0000553	Paxs	715.49	Joback Method
dvisc	0.0001168	Paxs	636.08	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369317&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369317&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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