

# 3'-Fluoroacetanilide

<b>Other names:</b>	3-Fluoroacetanilide m-Fluoroacetanilide Acetamide, N-(3-fluorophenyl)- Acetanilide, 3'-fluoro- N-(3-fluorophenyl)acetamide
<b>Inchi:</b>	InChI=1S/C8H8FNO/c1-6(11)10-8-4-2-3-7(9)5-8/h2-5H,1H3,(H,10,11)
<b>InchiKey:</b>	AQLLDCFUQXGLHM-UHFFFAOYSA-N
<b>Formula:</b>	C8H8FNO
<b>SMILES:</b>	CC(=O)Nc1cccc(F)c1
<b>Mol. weight [g/mol]:</b>	153.15
<b>CAS:</b>	351-28-0

## Physical Properties

Property code	Value	Unit	Source
chs	-4092.00	kJ/mol	NIST Webbook
gf	-115.08	kJ/mol	Joback Method
hf	-238.61	kJ/mol	Joback Method
hfus	19.91	kJ/mol	Joback Method
hvap	48.70	kJ/mol	Joback Method
log10ws	-2.01		Crippen Method
logp	1.784		Crippen Method
mcvol	113.140	ml/mol	McGowan Method
pc	3722.56	kPa	Joback Method
tb	517.41	K	Joback Method
tc	729.82	K	Joback Method
tf	322.04	K	Joback Method
vc	0.434	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	242.65	J/mol×K	517.41	Joback Method
cpg	253.72	J/mol×K	552.81	Joback Method
cpg	264.12	J/mol×K	588.21	Joback Method

cpg	273.86	J/mol×K	623.61	Joback Method
cpg	282.97	J/mol×K	659.01	Joback Method
cpg	291.48	J/mol×K	694.41	Joback Method
cpg	299.41	J/mol×K	729.82	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C351280&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C351280&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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