

# Dimethylmalonic acid, cis-4-methylcyclohexyl tridecyl ester

Inchi:	InChI=1S/C25H46O4/c1-5-6-7-8-9-10-11-12-13-14-15-20-28-23(26)25(3,4)24(27)29-22-1
InchiKey:	PWQUAZPRRYIBNE-UHFFFAOYSA-N
Formula:	C25H46O4
SMILES:	CCCCCCCCCCCCOC(=O)C(C)(C)C(=O)OC1CCC(C)CC1
Mol. weight [g/mol]:	410.63

## Physical Properties

Property code	Value	Unit	Source
gf	-288.64	kJ/mol	Joback Method
hf	-1023.70	kJ/mol	Joback Method
hfus	51.57	kJ/mol	Joback Method
hvap	88.38	kJ/mol	Joback Method
log10ws	-7.54		Crippen Method
logp	6.989		Crippen Method
mvol	367.130	ml/mol	McGowan Method
pc	899.10	kPa	Joback Method
rinpol	2742.00		NIST Webbook
rinpol	2742.00		NIST Webbook
tb	935.63	K	Joback Method
tc	1145.96	K	Joback Method
tf	521.39	K	Joback Method
vc	1.405	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1282.12	J/molxK	935.63	Joback Method
cpg	1365.81	J/molxK	1110.90	Joback Method
cpg	1352.08	J/molxK	1075.85	Joback Method
cpg	1336.90	J/molxK	1040.79	Joback Method
cpg	1320.22	J/molxK	1005.74	Joback Method
cpg	1301.98	J/molxK	970.68	Joback Method
cpg	1378.14	J/molxK	1145.96	Joback Method
dvisc	0.0000264	Paxs	935.63	Joback Method

dvisc	0.0000356	Paxs	866.59	Joback Method
dvisc	0.0000506	Paxs	797.55	Joback Method
dvisc	0.0000768	Paxs	728.51	Joback Method
dvisc	0.0001274	Paxs	659.47	Joback Method
dvisc	0.0002376	Paxs	590.43	Joback Method
dvisc	0.0005231	Paxs	521.39	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U363886&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U363886&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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