

# Alloaromadendrene epoxide

<b>Other names:</b>	Epoxy-allo-alloaromadendrene Alloaromadendrene oxide Alloaromadendrene, epoxy Epoxy-allo-aromadendrene
<b>Inchi:</b>	InChI=1S/C15H24O/c1-9-4-5-10-12(9)13-11(14(13,2)3)6-7-15(10)8-16-15/h9-13H,4-8H2
<b>InchiKey:</b>	XPGWKKLDFXNBPJ-PTYHTGKGS-A-N
<b>Formula:</b>	C15H24O
<b>SMILES:</b>	CC1CCC2C1C1C(CCC23CO3)C1(C)C
<b>Mol. weight [g/mol]:</b>	220.35
<b>CAS:</b>	85760-81-2

## Physical Properties

Property code	Value	Unit	Source
gf	186.09	kJ/mol	Joback Method
hf	-230.43	kJ/mol	Joback Method
hfus	23.64	kJ/mol	Joback Method
hvap	50.09	kJ/mol	Joback Method
log10ws	-3.43		Crippen Method
logp	3.484		Crippen Method
mcvol	184.640	ml/mol	McGowan Method
pc	2202.08	kPa	Joback Method
rinpol	1650.00		NIST Webbook
rinpol	1595.00		NIST Webbook
rinpol	1595.00		NIST Webbook
rinpol	1636.00		NIST Webbook
rinpol	1646.00		NIST Webbook
rinpol	1631.00		NIST Webbook
rinpol	1645.00		NIST Webbook
rinpol	1641.00		NIST Webbook
rinpol	1622.00		NIST Webbook
rinpol	1631.00		NIST Webbook
rinpol	1639.00		NIST Webbook
rinpol	1642.00		NIST Webbook
rinpol	1644.00		NIST Webbook
rinpol	1681.00		NIST Webbook
rinpol	1641.00		NIST Webbook
ripol	2095.00		NIST Webbook

ripol	2095.00		NIST Webbook
ripol	2054.00		NIST Webbook
tb	587.25	K	Joback Method
tc	815.53	K	Joback Method
tf	388.70	K	Joback Method
vc	0.710	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	544.52	J/mol×K	587.25	Joback Method
cpg	568.21	J/mol×K	625.30	Joback Method
cpg	590.27	J/mol×K	663.34	Joback Method
cpg	611.03	J/mol×K	701.39	Joback Method
cpg	630.84	J/mol×K	739.44	Joback Method
cpg	650.04	J/mol×K	777.48	Joback Method
cpg	668.97	J/mol×K	815.53	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C85760812&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C85760812&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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