

# Risocaine

<b>Other names:</b>	4-(Propoxycarbonyl)aniline Benzoic acid, 4-amino-, propyl ester Benzoic acid, p-amino-, propyl ester Keloform P NSC-23516 Propaesin Propazyl Propesin Propesine Propyl 4-aminobenzoate Propyl-p-amino benzoate Propylcain Raythesin n-Propyl p-aminobenzoate n-Propyl-4-aminobenzoate p-Aminobenzoic acid, propyl ester propyl-p-aminobenzoate
<b>Inchi:</b>	InChI=1S/C10H13NO2/c1-2-7-13-10(12)8-3-5-9(11)6-4-8/h3-6H,2,7,11H2,1H3
<b>InchiKey:</b>	NBFQYHKHPBMJJV-UHFFFAOYSA-N
<b>Formula:</b>	C10H13NO2
<b>SMILES:</b>	CCCOC(=O)c1ccc(N)cc1
<b>Mol. weight [g/mol]:</b>	179.22
<b>CAS:</b>	94-12-2

## Physical Properties

Property code	Value	Unit	Source
gf	-31.37	kJ/mol	Joback Method
hf	-235.68	kJ/mol	Joback Method
hfus	23.29	kJ/mol	Joback Method
hvap	60.59	kJ/mol	Joback Method
log10ws	-2.45		Estimated Solubility Method
log10ws	-2.33		Aqueous Solubility Prediction Method
logp	1.836		Crippen Method
mcvol	145.420	ml/mol	McGowan Method
pc	3231.98	kPa	Joback Method

rinpol	1659.00		NIST Webbook
rinpol	1731.00		NIST Webbook
rinpol	1670.00		NIST Webbook
tb	608.68	K	Joback Method
tc	830.79	K	Joback Method
tf	347.65	K	Aqueous Solubility Prediction Method
tf	346.00 ± 1.00	K	NIST Webbook
tf	347.00 ± 0.50	K	NIST Webbook
tf	345.80 ± 0.50	K	NIST Webbook
tf	346.00 ± 0.50	K	NIST Webbook
vc	0.540	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	357.09	J/mol×K	608.68	Joback Method
cpg	370.13	J/mol×K	645.70	Joback Method
cpg	382.38	J/mol×K	682.72	Joback Method
cpg	393.85	J/mol×K	719.74	Joback Method
cpg	404.56	J/mol×K	756.76	Joback Method
cpg	414.52	J/mol×K	793.78	Joback Method
cpg	423.76	J/mol×K	830.79	Joback Method
hfust	20.54	kJ/mol	347.10	NIST Webbook
hfust	20.54	kJ/mol	347.10	NIST Webbook
sfust	64.61	J/mol×K	347.10	NIST Webbook

## Sources

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C94122&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**Aqueous Solubility Prediction Method:** <http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

**Estimated Solubility Method:** [http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl\\_file/ci034243xsi20040112\\_053635.txt](http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt)

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/28-181-0/Risocaine.pdf>

Generated by Cheméo on 2024-04-20 08:03:19.446615304 +0000 UTC m=+15889448.367192619.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.