

# Heptane, 2,2,4-trimethyl-

<b>Other names:</b>	2,2,4-Trimethylheptane
<b>Inchi:</b>	InChI=1S/C10H22/c1-6-7-9(2)8-10(3,4)5/h9H,6-8H2,1-5H3
<b>InchiKey:</b>	IYGOARYARWJBO-UHFFFAOYSA-N
<b>Formula:</b>	C10H22
<b>SMILES:</b>	CCCC(C)CC(C)(C)C
<b>Mol. weight [g/mol]:</b>	142.28
<b>CAS:</b>	14720-74-2

## Physical Properties

Property code	Value	Unit	Source
gf	33.72	kJ/mol	Joback Method
hf	-263.76	kJ/mol	Joback Method
hfus	10.72	kJ/mol	Joback Method
hvap	45.60	kJ/mol	NIST Webbook
log10ws	-3.52		Crippen Method
logp	3.859		Crippen Method
mcvol	151.760	ml/mol	McGowan Method
pc	2161.32	kPa	Joback Method
rinpol	875.30		NIST Webbook
rinpol	900.00		NIST Webbook
rinpol	876.00		NIST Webbook
rinpol	875.00		NIST Webbook
rinpol	875.70		NIST Webbook
rinpol	889.00		NIST Webbook
rinpol	874.00		NIST Webbook
rinpol	876.00		NIST Webbook
rinpol	890.00		NIST Webbook
rinpol	884.00		NIST Webbook
rinpol	883.00		NIST Webbook
rinpol	876.00		NIST Webbook
rinpol	876.00		NIST Webbook
rinpol	890.00		NIST Webbook
rinpol	884.00		NIST Webbook
rinpol	875.00		NIST Webbook
rinpol	889.90		NIST Webbook
tb	425.12 ± 1.00	K	NIST Webbook
tb	422.55 ± 2.00	K	NIST Webbook

tb	419.15 ± 1.50	K	NIST Webbook
tc	600.93	K	Joback Method
tf	189.88	K	Joback Method
vc	0.579	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	316.93	J/mol×K	424.53	Joback Method
cpg	333.94	J/mol×K	453.93	Joback Method
cpg	350.18	J/mol×K	483.33	Joback Method
cpg	365.66	J/mol×K	512.73	Joback Method
cpg	380.41	J/mol×K	542.13	Joback Method
cpg	394.47	J/mol×K	571.53	Joback Method
cpg	407.85	J/mol×K	600.93	Joback Method
dvisc	0.0206027	Paxs	189.88	Joback Method
dvisc	0.0052013	Paxs	228.99	Joback Method
dvisc	0.0019620	Paxs	268.10	Joback Method
dvisc	0.0009486	Paxs	307.20	Joback Method
dvisc	0.0005405	Paxs	346.31	Joback Method
dvisc	0.0003452	Paxs	385.42	Joback Method
dvisc	0.0002394	Paxs	424.53	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.40223e+01
Coeff. B	-3.36840e+03
Coeff. C	-6.32730e+01
Temperature range (K), min.	308.52
Temperature range (K), max.	449.96

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$

Coeff. A	1.00470e+02
Coeff. B	-8.80083e+03
Coeff. C	-1.26277e+01
Coeff. D	7.57548e-06
Temperature range (K), min.	308.15
Temperature range (K), max.	594.50

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C14720742&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C14720742&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=127">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=127</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemo.com/doc/models/crippen_log10ws">https://www.chemo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=127">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=127</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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