

# Phenol, 2-bromo-, acetate

<b>Other names:</b>	Phenol, o-bromo-, acetate o-Acetoxybromobenzene o-Bromophenyl acetate 2-Bromophenol acetate Acetic acid, 2-bromophenyl ester 2-Bromophenyl acetate
<b>Inchi:</b>	InChI=1S/C8H7BrO2/c1-6(10)11-8-5-3-2-4-7(8)9/h2-5H,1H3
<b>InchiKey:</b>	BEHBHYYPTOHUHX-UHFFFAOYSA-N
<b>Formula:</b>	C8H7BrO2
<b>SMILES:</b>	CC(=O)Oc1ccccc1Br
<b>Mol. weight [g/mol]:</b>	215.04
<b>CAS:</b>	1829-37-4

## Physical Properties

Property code	Value	Unit	Source
gf	-100.34	kJ/mol	Joback Method
hf	-201.86	kJ/mol	Joback Method
hfus	18.20	kJ/mol	Joback Method
hvap	51.93	kJ/mol	Joback Method
ie	8.66 ± 0.03	eV	NIST Webbook
log10ws	-2.95		Crippen Method
logp	2.374		Crippen Method
mcpvol	124.760	ml/mol	McGowan Method
pc	4135.61	kPa	Joback Method
rinpol	1254.00		NIST Webbook
rinpol	1254.00		NIST Webbook
tb	556.55	K	Joback Method
tc	792.86	K	Joback Method
tf	350.82	K	Joback Method
vc	0.462	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	246.83	J/mol×K	556.55	Joback Method
cpg	291.72	J/mol×K	753.47	Joback Method
cpg	284.02	J/mol×K	714.09	Joback Method
cpg	275.70	J/mol×K	674.70	Joback Method
cpg	266.74	J/mol×K	635.32	Joback Method
cpg	257.12	J/mol×K	595.93	Joback Method
cpg	298.81	J/mol×K	792.86	Joback Method
dvisc	0.0002598	Paxs	556.55	Joback Method
dvisc	0.0003175	Paxs	522.26	Joback Method
dvisc	0.0003990	Paxs	487.97	Joback Method
dvisc	0.0005191	Paxs	453.69	Joback Method
dvisc	0.0007051	Paxs	419.40	Joback Method
dvisc	0.0010113	Paxs	385.11	Joback Method
dvisc	0.0015564	Paxs	350.82	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1829374&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1829374&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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