

# Hexane, 3,3,4-trimethyl-

<b>Other names:</b>	3,3,4-Trimethylhexane
<b>Inchi:</b>	InChI=1S/C9H20/c1-6-8(3)9(4,5)7-2/h8H,6-7H2,1-5H3
<b>InchiKey:</b>	ARWOOWBJJKVYOV-UHFFFAOYSA-N
<b>Formula:</b>	C9H20
<b>SMILES:</b>	CCC(C)C(C)(C)CC
<b>Mol. weight [g/mol]:</b>	128.26
<b>CAS:</b>	16747-31-2

## Physical Properties

Property code	Value	Unit	Source
chl	-6122.36 ± 0.75	kJ/mol	NIST Webbook
gf	25.30	kJ/mol	Joback Method
hf	-243.12	kJ/mol	Joback Method
hfl	-277.50 ± 0.92	kJ/mol	NIST Webbook
hfus	8.13	kJ/mol	Joback Method
hvap	42.20	kJ/mol	NIST Webbook
log10ws	-3.11		Crippen Method
logp	3.469		Crippen Method
mcvol	137.670	ml/mol	McGowan Method
pc	2367.97	kPa	Joback Method
rinpol	852.00		NIST Webbook
rinpol	855.00		NIST Webbook
rinpol	860.00		NIST Webbook
rinpol	851.70		NIST Webbook
rinpol	857.10		NIST Webbook
rinpol	850.80		NIST Webbook
rinpol	851.00		NIST Webbook
rinpol	855.10		NIST Webbook
rinpol	854.40		NIST Webbook
rinpol	837.00		NIST Webbook
rinpol	839.00		NIST Webbook
rinpol	839.00		NIST Webbook
rinpol	840.00		NIST Webbook
rinpol	841.00		NIST Webbook
rinpol	855.00		NIST Webbook
rinpol	847.00		NIST Webbook
rinpol	855.00		NIST Webbook

rinpol	858.00		NIST Webbook
rinpol	860.00		NIST Webbook
rinpol	854.00		NIST Webbook
rinpol	847.00		NIST Webbook
rinpol	847.00		NIST Webbook
rinpol	846.00		NIST Webbook
rinpol	849.00		NIST Webbook
rinpol	855.00		NIST Webbook
rinpol	855.00		NIST Webbook
rinpol	846.00		NIST Webbook
rinpol	855.00		NIST Webbook
rinpol	851.00		NIST Webbook
rinpol	839.00		NIST Webbook
rinpol	854.00		NIST Webbook
rinpol	848.00		NIST Webbook
ripol	816.00		NIST Webbook
ripol	820.00		NIST Webbook
ripol	816.00		NIST Webbook
tb	413.05 ± 0.50	K	NIST Webbook
tb	413.57 ± 0.15	K	NIST Webbook
tc	579.13	K	Joback Method
tf	171.95 ± 0.05	K	NIST Webbook
vc	0.522	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	273.89	J/molxK	401.65	Joback Method
cpg	289.98	J/molxK	431.23	Joback Method
cpg	305.33	J/molxK	460.81	Joback Method
cpg	319.96	J/molxK	490.39	Joback Method
cpg	333.90	J/molxK	519.97	Joback Method
cpg	347.18	J/molxK	549.55	Joback Method
cpg	359.82	J/molxK	579.13	Joback Method
dvisc	0.0222427	Paxs	178.61	Joback Method
dvisc	0.0055664	Paxs	215.78	Joback Method
dvisc	0.0020930	Paxs	252.96	Joback Method
dvisc	0.0010112	Paxs	290.13	Joback Method
dvisc	0.0005763	Paxs	327.30	Joback Method
dvisc	0.0003684	Paxs	364.48	Joback Method
dvisc	0.0002558	Paxs	401.65	Joback Method

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.38943e+01
Coeff. B	-3.23887e+03
Coeff. C	-6.44510e+01
Temperature range (K), min.	302.49
Temperature range (K), max.	441.82

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	8.53062e+01
Coeff. B	-7.82480e+03
Coeff. C	-1.04380e+01
Coeff. D	6.54368e-06
Temperature range (K), min.	300.15
Temperature range (K), max.	602.30

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=89">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=89</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C16747312&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C16747312&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=89">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=89</a>

# Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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