

# Hexyl iodoacetate

<b>Inchi:</b>	InChI=1S/C8H15IO2/c1-2-3-4-5-6-11-8(10)7-9/h2-7H2,1H3
<b>InchiKey:</b>	XJOFYNWBVZSJZ-UHFFFAOYSA-N
<b>Formula:</b>	C8H15IO2
<b>SMILES:</b>	CCCCCOC(=O)CI
<b>Mol. weight [g/mol]:</b>	270.11

## Physical Properties

Property code	Value	Unit	Source
gf	-159.32	kJ/mol	Joback Method
hf	-376.38	kJ/mol	Joback Method
hfus	23.67	kJ/mol	Joback Method
hvap	51.93	kJ/mol	Joback Method
log10ws	-2.98		Crippen Method
logp	2.545		Crippen Method
mcvol	156.840	ml/mol	McGowan Method
pc	2597.78	kPa	Joback Method
rinpol	1382.30		NIST Webbook
rinpol	1349.90		NIST Webbook
rinpol	1349.90		NIST Webbook
tb	551.87	K	Joback Method
tc	755.24	K	Joback Method
tf	310.14	K	Joback Method
vc	0.596	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	332.40	J/molxK	551.87	Joback Method
cpg	387.10	J/molxK	721.34	Joback Method
cpg	377.28	J/molxK	687.45	Joback Method
cpg	366.92	J/molxK	653.55	Joback Method
cpg	356.00	J/molxK	619.66	Joback Method
cpg	344.49	J/molxK	585.76	Joback Method
cpg	396.37	J/molxK	755.24	Joback Method

dvisc	0.0002646	Paxs	551.87	Joback Method
dvisc	0.0003415	Paxs	511.58	Joback Method
dvisc	0.0004603	Paxs	471.29	Joback Method
dvisc	0.0006562	Paxs	431.00	Joback Method
dvisc	0.0010063	Paxs	390.72	Joback Method
dvisc	0.0017028	Paxs	350.43	Joback Method
dvisc	0.0033031	Paxs	310.14	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R248283&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R248283&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/28-279-2/Hexyl-iodoacetate.pdf>

Generated by Cheméo on 2024-05-01 13:20:19.430217987 +0000 UTC m=+16858868.350795309.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.