

# Pentafluoropropionic acid, pentyl ester

<b>Other names:</b>	Pentyl pentafluoropropionate Pentyl 2,2,3,3,3-pentafluoropropanoate 1-Pentanol, pentafluoropropionate 2,2,3,3,3-Pentafluoro-propionic acid pentyl ester Pentyl pentafluoropropanoate Propanoic acid, pentafluoro, pentyl ester
<b>Inchi:</b>	InChI=1S/C8H11F5O2/c1-2-3-4-5-15-6(14)7(9,10)8(11,12)13/h2-5H2,1H3
<b>InchiKey:</b>	BPYYNYLCDRAOKR-UHFFFAOYSA-N
<b>Formula:</b>	C8H11F5O2
<b>SMILES:</b>	CCCCCOC(=O)C(F)(F)C(F)(F)F
<b>Mol. weight [g/mol]:</b>	234.16
<b>CAS:</b>	680-29-5

## Physical Properties

Property code	Value	Unit	Source
gf	-1185.81	kJ/mol	Joback Method
hf	-1451.30	kJ/mol	Joback Method
hfus	19.84	kJ/mol	Joback Method
hvap	35.88	kJ/mol	Joback Method
log10ws	-3.01		Crippen Method
logp	2.917		Crippen Method
mcvol	139.870	ml/mol	McGowan Method
pc	2210.37	kPa	Joback Method
rinpol	780.70		NIST Webbook
rinpol	784.00		NIST Webbook
rinpol	798.00		NIST Webbook
rinpol	781.00		NIST Webbook
rinpol	802.50		NIST Webbook
ripol	820.00		NIST Webbook
tb	448.62	K	Joback Method
tc	602.67	K	Joback Method
tf	259.87	K	Joback Method
vc	0.576	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	326.06	J/mol×K	448.62	Joback Method
cpg	337.69	J/mol×K	474.29	Joback Method
cpg	348.75	J/mol×K	499.97	Joback Method
cpg	359.26	J/mol×K	525.64	Joback Method
cpg	369.23	J/mol×K	551.32	Joback Method
cpg	378.69	J/mol×K	576.99	Joback Method
cpg	387.66	J/mol×K	602.67	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C680295&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C680295&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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