

# Benzaldehyde, 3-chloro-

<b>Other names:</b>	Benzaldehyde, m-chloro- m-Chlorobenzaldehyde meta-Chlorobenzaldehyde 3-Chlorobenzaldehyde
<b>Inchi:</b>	InChI=1S/C7H5ClO/c8-7-3-1-2-6(4-7)5-9/h1-5H
<b>InchiKey:</b>	SRWILAKSARHZPR-UHFFFAOYSA-N
<b>Formula:</b>	C7H5ClO
<b>SMILES:</b>	O=Cc1cccc(Cl)c1
<b>Mol. weight [g/mol]:</b>	140.57
<b>CAS:</b>	587-04-2

## Physical Properties

Property code	Value	Unit	Source
affp	813.00	kJ/mol	NIST Webbook
basg	781.10	kJ/mol	NIST Webbook
chl	-3366.90 ± 8.40	kJ/mol	NIST Webbook
chl	-3365.00	kJ/mol	NIST Webbook
ea	0.71 ± 0.09	eV	NIST Webbook
ea	0.67 ± 0.09	eV	NIST Webbook
gf	-0.61	kJ/mol	Joback Method
hf	-64.07	kJ/mol	Joback Method
hfl	-105.00	kJ/mol	NIST Webbook
hfus	14.02	kJ/mol	Joback Method
hvap	45.22	kJ/mol	Joback Method
log10ws	-2.40		Crippen Method
logp	2.152		Crippen Method
mcvol	99.540	ml/mol	McGowan Method
pc	4233.04	kPa	Joback Method
rinpol	1102.00		NIST Webbook
rinpol	1118.90		NIST Webbook
rinpol	1108.40		NIST Webbook
tb	486.70	K	NIST Webbook
tc	704.10	K	Joback Method
tf	279.51	K	Joback Method
vc	0.386	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	177.88	J/molxK	477.31	Joback Method
cpg	186.90	J/molxK	515.11	Joback Method
cpg	195.31	J/molxK	552.91	Joback Method
cpg	203.14	J/molxK	590.70	Joback Method
cpg	210.42	J/molxK	628.50	Joback Method
cpg	217.17	J/molxK	666.30	Joback Method
cpg	223.42	J/molxK	704.10	Joback Method
dvisc	0.0023338	Paxs	279.51	Joback Method
dvisc	0.0014204	Paxs	312.48	Joback Method
dvisc	0.0009504	Paxs	345.44	Joback Method
dvisc	0.0006820	Paxs	378.41	Joback Method
dvisc	0.0005162	Paxs	411.38	Joback Method
dvisc	0.0004071	Paxs	444.34	Joback Method
dvisc	0.0003318	Paxs	477.31	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C587042&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C587042&amp;Units=SI</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions

<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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