

# Trimethyl orthobutyrates

<b>Other names:</b>	ortho-n-Butyric acid trimethyl ester Butane, 1,1,1-trimethoxy- 1,1,1-trimethoxybutane
<b>Inchi:</b>	InChI=1S/C7H16O3/c1-5-6-7(8-2,9-3)10-4/h5-6H2,1-4H3
<b>InchiKey:</b>	JAFMOTJMRSZOJE-UHFFFAOYSA-N
<b>Formula:</b>	C7H16O3
<b>SMILES:</b>	CCCC(OC)(OC)OC
<b>Mol. weight [g/mol]:</b>	148.20
<b>CAS:</b>	43083-12-1

## Physical Properties

Property code	Value	Unit	Source
gf	-304.10	kJ/mol	Joback Method
hf	-593.22	kJ/mol	Joback Method
hfus	10.04	kJ/mol	Joback Method
hvap	37.11	kJ/mol	Joback Method
log10ws	-1.12		Crippen Method
logp	1.380		Crippen Method
mcvol	127.100	ml/mol	McGowan Method
pc	2707.03	kPa	Joback Method
tb	419.20	K	NIST Webbook
tc	599.04	K	Joback Method
tf	237.76	K	Joback Method
vc	0.470	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	268.18	J/molxK	423.59	Joback Method
cpg	280.51	J/molxK	452.83	Joback Method
cpg	292.46	J/molxK	482.07	Joback Method
cpg	304.02	J/molxK	511.32	Joback Method
cpg	315.20	J/molxK	540.56	Joback Method
cpg	325.98	J/molxK	569.80	Joback Method

cpg	336.37	J/mol×K	599.04	Joback Method
dvisc	0.0033515	Paxs	237.76	Joback Method
dvisc	0.0015609	Paxs	268.73	Joback Method
dvisc	0.0008514	Paxs	299.70	Joback Method
dvisc	0.0005202	Paxs	330.67	Joback Method
dvisc	0.0003458	Paxs	361.65	Joback Method
dvisc	0.0002452	Paxs	392.62	Joback Method
dvisc	0.0001828	Paxs	423.59	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C43083121&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C43083121&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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