

2,6-Diallyl-4-chlorophenol

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|-----------------------------|---|
| Inchi: | InChI=1S/C12H13ClO/c1-3-5-9-7-11(13)8-10(6-4-2)12(9)14/h3-4,7-8,14H,1-2,5-6H2 |
| InchiKey: | RANSQHWYECVCMB-UHFFFAOYSA-N |
| Formula: | C12H13ClO |
| SMILES: | <chem>C=CCc1cc(Cl)cc(CC=C)c1O</chem> |
| Mol. weight [g/mol]: | 208.68 |
| CAS: | 73267-76-2 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | 152.44 | kJ/mol | Joback Method |
| hf | -19.61 | kJ/mol | Joback Method |
| hfus | 27.52 | kJ/mol | Joback Method |
| hvap | 61.97 | kJ/mol | Joback Method |
| log10ws | -3.85 | | Crippen Method |
| logp | 3.503 | | Crippen Method |
| mvol | 165.690 | ml/mol | McGowan Method |
| pc | 2850.52 | kPa | Joback Method |
| tb | 622.01 | K | Joback Method |
| tc | 847.72 | K | Joback Method |
| tf | 414.58 | K | Joback Method |
| vc | 0.577 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 388.58 | J/molxK | 622.01 | Joback Method |
| cpg | 443.42 | J/molxK | 810.10 | Joback Method |
| cpg | 433.71 | J/molxK | 772.48 | Joback Method |
| cpg | 423.46 | J/molxK | 734.86 | Joback Method |
| cpg | 412.58 | J/molxK | 697.25 | Joback Method |
| cpg | 400.98 | J/molxK | 659.63 | Joback Method |
| cpg | 452.68 | J/molxK | 847.72 | Joback Method |
| dvisc | 0.0000306 | Paxs | 622.01 | Joback Method |
| dvisc | 0.0000443 | Paxs | 587.44 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0000673 | Paxs | 552.87 | Joback Method |
| dvisc | 0.0001081 | Paxs | 518.30 | Joback Method |
| dvisc | 0.0001858 | Paxs | 483.72 | Joback Method |
| dvisc | 0.0003469 | Paxs | 449.15 | Joback Method |
| dvisc | 0.0007190 | Paxs | 414.58 | Joback Method |

Sources

| | |
|------------------------|---|
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C73267762&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mccvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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