2,3,4-Trimethyl-1-pentanol

Inchi: InChI=1S/C8H18O/c1-6(2)8(4)7(3)5-9/h6-9H,5H2,1-4H3
InchiKey: PAZDSSMTPLLLIR-UHFFFAOYSA-N
Formula: C8H18O
SMILES: CC(C)C(C)C(C)CO
Mol. weight [g/mol]: 130.23
CAS: 6570-88-3

Physical Properties

Property code	Value	Unit	Source
gf	-127.66	kJ/mol	Joback Method
hf	-376.52	kJ/mol	Joback Method
hfus	9.99	kJ/mol	Joback Method
hvap	48.92	kJ/mol	Joback Method
log10ws	-1.71		Crippen Method
logp	1.907		Crippen Method
mcvol	129.450	ml/mol	McGowan Method
pc	2859.68	kPa	Joback Method
tb	473.30	K	Joback Method
tc	642.89	K	Joback Method
tf	195.74	K	Joback Method
vc	0.484	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	290.54	J/mol×K	473.30	Joback Method
cpg	303.02	J/mol×K	501.56	Joback Method
cpg	314.99	J/mol×K	529.83	Joback Method
cpg	326.48	J/mol×K	558.09	Joback Method
cpg	337.49	J/mol×K	586.36	Joback Method
cpg	348.04	J/mol×K	614.62	Joback Method
cpg	358.15	J/mol×K	642.89	Joback Method
dvisc	0.9006667	Pa×s	195.74	Joback Method
dvisc	0.0532070	Pa×s	242.00	Joback Method

dvisc	0.0077930	Pa×s	288.26	Joback Method
dvisc	0.0019417	Pa×s	334.52	Joback Method
dvisc	0.0006781	Pa×s	380.78	Joback Method
dvisc	0.0002974	Pa×s	427.04	Joback Method
dvisc	0.0001533	Pa×s	473.30	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	ln(Pvp) = A + B/(T + C)
Coeff. A	1.87397e+01
Coeff. B	-6.26934e+03
Coeff. C	-1.50000e-01
Temperature range (K), min.	339.91
Temperature range (K), max.	467.03

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C6570883&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l

logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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