

# Glutaric acid monoamide, N-(1,2,3,4-tetrahydronaphth-1-yl)-, propyl ester

Other names:

Glutaric acid, N-(1,2,3,4-tetrahydronaphth-1-yl)-, propyl ester

Inchi:

InChI=1S/C18H25NO3/c1-2-13-22-18(21)12-6-11-17(20)19-16-10-5-8-14-7-3-4-9-15(14)

InchiKey:

RAMPUCPUBCAYHC-UHFFFAOYSA-N

Formula:

C18H25NO3

SMILES:

CCCOC(=O)CCCC(=O)NC1CCc2ccccc21

Mol. weight [g/mol]:

303.40

## Physical Properties

Property code	Value	Unit	Source
gf	-21.34	kJ/mol	Joback Method
hf	-427.06	kJ/mol	Joback Method
hfus	41.55	kJ/mol	Joback Method
hvap	81.02	kJ/mol	Joback Method
log10ws	-4.61		Crippen Method
logp	3.304		Crippen Method
mvol	248.850	ml/mol	McGowan Method
pc	1806.16	kPa	Joback Method
rinpol	2560.00		NIST Webbook
rinpol	2560.00		NIST Webbook
tb	834.24	K	Joback Method
tc	1049.49	K	Joback Method
tf	520.73	K	Joback Method
vc	0.950	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	777.66	J/mol×K	834.24	Joback Method
cpg	793.42	J/mol×K	870.11	Joback Method
cpg	808.03	J/mol×K	905.99	Joback Method
cpg	821.57	J/mol×K	941.86	Joback Method
cpg	834.07	J/mol×K	977.74	Joback Method
cpg	845.61	J/mol×K	1013.61	Joback Method
cpg	856.24	J/mol×K	1049.49	Joback Method

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U360203&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U360203&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/28-342-1/Glutaric-acid-monoamide-N-1-2-3-4-tetrahydronaphth-1-yl-propyl-ester.pdf>

Generated by Cheméo on 2024-04-19 17:30:12.584199571 +0000 UTC m=+15837061.504776893.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.