

# trans-2-Decenoic acid

<b>Other names:</b>	2-Decenoic acid, (E)- E-2-Decenoic acid
<b>Inchi:</b>	InChI=1S/C10H18O2/c1-2-3-4-5-6-7-8-9-10(11)12/h8-9H,2-7H2,1H3,(H,11,12)/b9-8+
<b>InchiKey:</b>	WXBXVVIUZANZAU-CMDGGOBGSA-N
<b>Formula:</b>	C10H18O2
<b>SMILES:</b>	CCCCCCCC=CC(=O)O
<b>Mol. weight [g/mol]:</b>	170.25
<b>CAS:</b>	334-49-6

## Physical Properties

Property code	Value	Unit	Source
gf	-152.20	kJ/mol	Joback Method
hf	-397.32	kJ/mol	Joback Method
hfus	27.55	kJ/mol	Joback Method
hvap	61.24	kJ/mol	Joback Method
log10ws	-2.96		Crippen Method
logp	2.988		Crippen Method
mcvol	154.900	ml/mol	McGowan Method
pc	2574.11	kPa	Joback Method
rinpol	1392.00		NIST Webbook
rinpol	1392.00		NIST Webbook
ripol	2428.00		NIST Webbook
ripol	2428.00		NIST Webbook
tb	578.41	K	Joback Method
tc	751.50	K	Joback Method
tf	308.13	K	Joback Method
vc	0.601	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	383.20	J/molxK	578.41	Joback Method
cpg	395.15	J/molxK	607.26	Joback Method
cpg	406.55	J/molxK	636.11	Joback Method

cpg	417.42	J/mol×K	664.96	Joback Method
cpg	427.78	J/mol×K	693.80	Joback Method
cpg	437.66	J/mol×K	722.65	Joback Method
cpg	447.07	J/mol×K	751.50	Joback Method
dvisc	0.0110783	Paxs	308.13	Joback Method
dvisc	0.0029530	Paxs	353.18	Joback Method
dvisc	0.0010616	Paxs	398.22	Joback Method
dvisc	0.0004699	Paxs	443.27	Joback Method
dvisc	0.0002417	Paxs	488.32	Joback Method
dvisc	0.0001391	Paxs	533.36	Joback Method
dvisc	0.0000873	Paxs	578.41	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.38286e+01
Coeff. B	-4.35701e+03
Coeff. C	-9.07560e+01
Temperature range (K), min.	412.52
Temperature range (K), max.	602.31

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C334496&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C334496&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity

<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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