

# Oxazole

<b>Other names:</b>	1,3-OXAZOLE
<b>Inchi:</b>	InChI=1S/C3H3NO/c1-2-5-3-4-1/h1-3H
<b>InchiKey:</b>	ZCQWOFVYLHDMMC-UHFFFAOYSA-N
<b>Formula:</b>	C3H3NO
<b>SMILES:</b>	c1cocn1
<b>Mol. weight [g/mol]:</b>	69.06
<b>CAS:</b>	288-42-6

## Physical Properties

Property code	Value	Unit	Source
affp	876.40	kJ/mol	NIST Webbook
basg	844.50	kJ/mol	NIST Webbook
chl	-1561.30 ± 0.50	kJ/mol	NIST Webbook
hf	-15.50 ± 0.54	kJ/mol	NIST Webbook
hfl	-48.03 ± 0.54	kJ/mol	NIST Webbook
hvap	32.50 ± 0.10	kJ/mol	NIST Webbook
hvap	32.50 ± 0.10	kJ/mol	NIST Webbook
ie	10.15	eV	NIST Webbook
ie	9.60	eV	NIST Webbook
ie	9.90	eV	NIST Webbook
log10ws	-4.97		Crippen Method
logp	0.675		Crippen Method
mcvol	49.520	ml/mol	McGowan Method
pc	6770.00	kPa	Critical Point and Vapor Pressure Measurements for 17 Compounds by a Low Residence Time Flow Method
rinpol	580.00		NIST Webbook
rinpol	97.30		NIST Webbook
rinpol	97.30		NIST Webbook
rinpol	580.00		NIST Webbook
rinpol	580.00		NIST Webbook
rinpol	538.00		NIST Webbook
rinpol	580.00		NIST Webbook
ripol	1019.00		NIST Webbook
ripol	1019.00		NIST Webbook
ripol	1019.00		NIST Webbook

tb	342.71 ± 0.20	K	NIST Webbook
tb	342.70	K	NIST Webbook

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hvapt	34.60	kJ/mol	318.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.47512e+01
Coeff. B	-3.11216e+03
Coeff. C	-3.55670e+01
Temperature range (K), min.	250.74
Temperature range (K), max.	365.25

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Critical Point and Vapor Pressure Measurements for 17 Compounds by a KDF Residence Time Flow Method:</b>	<a href="https://www.doi.org/10.1021/je060269j">https://www.doi.org/10.1021/je060269j</a>
<b>McGowan Method:</b>	<a href="https://www.thermochimica.com/files/research/kdb/mol/mol1461.mol">https://www.thermochimica.com/files/research/kdb/mol/mol1461.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C288426&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C288426&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

**affp:** Proton affinity

<b>basg:</b>	Gas basicity
<b>chl:</b>	Standard liquid enthalpy of combustion
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

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