

# 4-Hexadecyloxybenzoic acid

|                      |                                                                                    |
|----------------------|------------------------------------------------------------------------------------|
| Other names:         | p-Hexadecyloxybenzoic acid<br>Benzoic acid, 4-(hexadecyloxy)-                      |
| Inchi:               | InChI=1S/C23H38O3/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-20-26-22-18-16-21(17-19-22) |
| InchiKey:            | GAQCVRTXIAGNEM-UHFFFAOYSA-N                                                        |
| Formula:             | C23H38O3                                                                           |
| SMILES:              | CCCCCCCCCCCCCCCCOc1ccc(C(=O)O)cc1                                                  |
| Mol. weight [g/mol]: | 362.55                                                                             |
| CAS:                 | 15872-48-7                                                                         |

## Physical Properties

| Property code | Value         | Unit    | Source         |
|---------------|---------------|---------|----------------|
| gf            | -125.18       | kJ/mol  | Joback Method  |
| hf            | -690.02       | kJ/mol  | Joback Method  |
| hfus          | 55.85         | kJ/mol  | Joback Method  |
| hvap          | 95.56         | kJ/mol  | Joback Method  |
| log10ws       | -7.93         |         | Crippen Method |
| logp          | 7.245         |         | Crippen Method |
| mcvol         | 324.480       | ml/mol  | McGowan Method |
| pc            | 1132.15       | kPa     | Joback Method  |
| tb            | 925.77        | K       | Joback Method  |
| tc            | 1133.44       | K       | Joback Method  |
| tf            | 375.00 ± 1.00 | K       | NIST Webbook   |
| vc            | 1.258         | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value   | Unit    | Temperature [K] | Source        |
|---------------|---------|---------|-----------------|---------------|
| cpg           | 1075.54 | J/mol×K | 925.77          | Joback Method |
| cpg           | 1092.53 | J/mol×K | 960.38          | Joback Method |
| cpg           | 1108.35 | J/mol×K | 994.99          | Joback Method |
| cpg           | 1123.05 | J/mol×K | 1029.61         | Joback Method |
| cpg           | 1136.67 | J/mol×K | 1064.22         | Joback Method |
| cpg           | 1149.28 | J/mol×K | 1098.83         | Joback Method |
| cpg           | 1160.93 | J/mol×K | 1133.44         | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0003152 | Pa×s | 520.89 | Joback Method |
| dvisc | 0.0001159 | Pa×s | 588.37 | Joback Method |
| dvisc | 0.0000523 | Pa×s | 655.85 | Joback Method |
| dvisc | 0.0000274 | Pa×s | 723.33 | Joback Method |
| dvisc | 0.0000160 | Pa×s | 790.81 | Joback Method |
| dvisc | 0.0000102 | Pa×s | 858.29 | Joback Method |
| dvisc | 0.0000069 | Pa×s | 925.77 | Joback Method |

## Sources

|                        |                                                                                                                                               |
|------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------|
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                         |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C15872487&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C15872487&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                     |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                             |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                         |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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