

# 4-Methyl-3-nitrophenol

Other names:	2-Nitro-4-hydroxytoluene 3-Nitro-4-methylphenol 3-Nitro-p-cresol 4-Hydroxy-2-nitrotoluene Phenol, 4-methyl-3-nitro- p-Cresol, 3-nitro-
Inchi:	InChI=1S/C7H7NO3/c1-5-2-3-6(9)4-7(5)8(10)11/h2-4,9H,1H3
InchiKey:	BQEXDUKMTVYBRK-UHFFFAOYSA-N
Formula:	C7H7NO3
SMILES:	Cc1ccc(O)cc1[N+](=O)[O-]
Mol. weight [g/mol]:	153.14
CAS:	2042-14-0

## Physical Properties

Property code	Value	Unit	Source
gf	-8.23	kJ/mol	Joback Method
hf	-150.82	kJ/mol	Joback Method
hfus	24.68	kJ/mol	Joback Method
hvap	63.72	kJ/mol	Joback Method
log10ws	-2.15		Crippen Method
logp	1.609		Crippen Method
mcvol	109.020	ml/mol	McGowan Method
pc	5008.59	kPa	Joback Method
tb	623.68	K	Joback Method
tc	883.67	K	Joback Method
tf	462.92	K	Joback Method
vc	0.367	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	262.28	J/mol×K	623.68	Joback Method
cpg	271.81	J/mol×K	667.01	Joback Method
cpg	280.56	J/mol×K	710.34	Joback Method

cpg	288.66	J/mol×K	753.67	Joback Method
cpg	296.21	J/mol×K	797.01	Joback Method
cpg	303.32	J/mol×K	840.34	Joback Method
cpg	310.10	J/mol×K	883.67	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{\text{vap}}) = A + B/(T + C)$
Coeff. A	1.01200e+01
Coeff. B	-2.50274e+03
Coeff. C	-8.42820e+01
Temperature range (K), min.	338.82
Temperature range (K), max.	604.76

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2042140&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2042140&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logP:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/28-572-6/4-Methyl-3-nitrophenol.pdf>

Generated by Cheméo on 2024-04-09 10:49:24.833679441 +0000 UTC m=+14949013.754256762.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.