

# Cyclohexanol, 1-propyl-

<b>Other names:</b>	1-Propylcyclohexanol 1-n-Propylcyclohexanol
<b>Inchi:</b>	InChI=1S/C9H18O/c1-2-6-9(10)7-4-3-5-8-9/h10H,2-8H2,1H3
<b>InchiKey:</b>	PYLPYOPJKOJRNP-UHFFFAOYSA-N
<b>Formula:</b>	C9H18O
<b>SMILES:</b>	CCCC1(O)CCCCC1
<b>Mol. weight [g/mol]:</b>	142.24
<b>CAS:</b>	5445-24-9

## Physical Properties

Property code	Value	Unit	Source
gf	-92.96	kJ/mol	Joback Method
hf	-311.76	kJ/mol	Joback Method
hfus	8.69	kJ/mol	Joback Method
hvap	51.59	kJ/mol	Joback Method
log10ws	-2.86		Crippen Method
logp	2.482		Crippen Method
mvol	132.680	ml/mol	McGowan Method
pc	3302.95	kPa	Joback Method
tb	453.00 ± 4.00	K	NIST Webbook
tc	713.09	K	Joback Method
tf	283.29	K	Joback Method
vc	0.489	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	320.65	J/mol×K	517.29	Joback Method
cpg	336.17	J/mol×K	549.92	Joback Method
cpg	350.77	J/mol×K	582.56	Joback Method
cpg	364.56	J/mol×K	615.19	Joback Method
cpg	377.60	J/mol×K	647.82	Joback Method
cpg	389.99	J/mol×K	680.46	Joback Method
cpg	401.82	J/mol×K	713.09	Joback Method

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.38284e+01
Coeff. B	-3.82131e+03
Coeff. C	-6.80910e+01
Temperature range (K), min.	350.30
Temperature range (K), max.	516.76

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5445249&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5445249&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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