

# 10,14-Dimethylpentadecyl isobutyrate

<b>Inchi:</b>	InChI=1S/C21H42O2/c1-18(2)14-13-16-20(5)15-11-9-7-6-8-10-12-17-23-21(22)19(3)4/h1
<b>InchiKey:</b>	WEUUVFYXSHQOIT-UHFFFAOYSA-N
<b>Formula:</b>	C21H42O2
<b>SMILES:</b>	CC(C)CCCC(C)CCCCCCCCCOC(=O)C(C)C
<b>Mol. weight [g/mol]:</b>	326.56

## Physical Properties

Property code	Value	Unit	Source
gf	-115.30	kJ/mol	Joback Method
hf	-737.41	kJ/mol	Joback Method
hfus	42.36	kJ/mol	Joback Method
hvap	70.33	kJ/mol	Joback Method
log10ws	-6.75		Crippen Method
logp	6.769		Crippen Method
mcvol	314.190	ml/mol	McGowan Method
pc	998.91	kPa	Joback Method
rinpol	2147.00		NIST Webbook
tb	754.85	K	Joback Method
tc	931.67	K	Joback Method
tf	353.59	K	Joback Method
vc	1.218	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	970.62	J/molxK	754.85	Joback Method
cpg	1063.27	J/molxK	902.20	Joback Method
cpg	1046.69	J/molxK	872.73	Joback Method
cpg	1029.16	J/molxK	843.26	Joback Method
cpg	1010.65	J/molxK	813.79	Joback Method
cpg	991.15	J/molxK	784.32	Joback Method
cpg	1078.94	J/molxK	931.67	Joback Method
dvisc	0.0000501	Paxs	754.85	Joback Method
dvisc	0.0000715	Paxs	687.97	Joback Method

dvisc	0.0001103	Paxs	621.10	Joback Method
dvisc	0.0001889	Paxs	554.22	Joback Method
dvisc	0.0003751	Paxs	487.34	Joback Method
dvisc	0.0009261	Paxs	420.47	Joback Method
dvisc	0.0032188	Paxs	353.59	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R422783&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R422783&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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