

Dimethylmalonic acid, hexadecyl 2,2,3,3-tetrafluoropropyl ester

Inchi: InChI=1S/C24H42F4O4/c1-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-31-21(29)23(2,3)22
InchiKey: ZBQYVJXYLLYKHC-UHFFFAOYSA-N
Formula: C24H42F4O4
SMILES: CCCCCCCCCCCCCCCCOC(=O)C(C)(C)C(=O)OCC(F)(F)C(F)F
Mol. weight [g/mol]: 470.58

Physical Properties

Property code	Value	Unit	Source
gf	-1092.64	kJ/mol	Joback Method
hf	-1835.51	kJ/mol	Joback Method
hfus	57.46	kJ/mol	Joback Method
hvap	81.08	kJ/mol	Joback Method
log10ws	-7.98		Crippen Method
logp	7.481		Crippen Method
mvol	370.980	ml/mol	McGowan Method
pc	787.27	kPa	Joback Method
rinpol	2403.00		NIST Webbook
rinpol	2403.00		NIST Webbook
tb	891.28	K	Joback Method
tc	1094.01	K	Joback Method
tf	496.76	K	Joback Method
vc	1.472	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1246.51	J/mol×K	891.28	Joback Method
cpg	1265.88	J/mol×K	925.07	Joback Method
cpg	1283.95	J/mol×K	958.86	Joback Method
cpg	1300.79	J/mol×K	992.65	Joback Method
cpg	1316.49	J/mol×K	1026.43	Joback Method
cpg	1331.10	J/mol×K	1060.22	Joback Method
cpg	1344.71	J/mol×K	1094.01	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U361926&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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