

# 4-Bromobutyric acid, 4-benzyloxyphenyl ester

<b>Inchi:</b>	InChI=1S/C17H17BrO3/c18-12-4-7-17(19)21-16-10-8-15(9-11-16)20-13-14-5-2-1-3-6-14
<b>InchiKey:</b>	NDYIZAFRVHKCHC-UHFFFAOYSA-N
<b>Formula:</b>	C17H17BrO3
<b>SMILES:</b>	O=C(CCCBr)Oc1ccc(OCc2ccccc2)cc1
<b>Mol. weight [g/mol]:</b>	349.22

## Physical Properties

Property code	Value	Unit	Source
gf	-17.15	kJ/mol	Joback Method
hf	-283.31	kJ/mol	Joback Method
hfus	36.74	kJ/mol	Joback Method
hvap	76.65	kJ/mol	Joback Method
log10ws	-5.28		Crippen Method
logp	4.346		Crippen Method
mvol	233.680	ml/mol	McGowan Method
pc	2256.81	kPa	Joback Method
rinpol	2568.00		NIST Webbook
rinpol	2568.00		NIST Webbook
tb	811.57	K	Joback Method
tc	1047.65	K	Joback Method
tf	500.90	K	Joback Method
vc	0.875	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	635.62	J/molxK	811.57	Joback Method
cpg	693.25	J/molxK	1008.30	Joback Method
cpg	683.94	J/molxK	968.96	Joback Method
cpg	673.57	J/molxK	929.61	Joback Method
cpg	662.09	J/molxK	890.26	Joback Method
cpg	649.46	J/molxK	850.92	Joback Method
cpg	701.56	J/molxK	1047.65	Joback Method
dvisc	0.0000708	Paxs	811.57	Joback Method

dvisc	0.0000889	Paxs	759.79	Joback Method
dvisc	0.0001153	Paxs	708.01	Joback Method
dvisc	0.0001558	Paxs	656.24	Joback Method
dvisc	0.0002217	Paxs	604.46	Joback Method
dvisc	0.0003370	Paxs	552.68	Joback Method
dvisc	0.0005585	Paxs	500.90	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U307614&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U307614&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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