

# 2-isopropyladamantane

**Inchi:** InChI=1S/C13H22/c1-8(2)13-11-4-9-3-10(6-11)7-12(13)5-9/h8-13H,3-7H2,1-2H3/t9-,10?,  
**InchiKey:** GNFZEYQQRJBGMB-VNAIHZQWSA-N  
**Formula:** C13H22  
**SMILES:** CC(C)C1C2CC3CC(C2)CC1C3  
**Mol. weight [g/mol]:** 178.31

## Physical Properties

Property code	Value	Unit	Source
gf	210.87	kJ/mol	Joback Method
hf	-145.37	kJ/mol	Joback Method
hfus	20.35	kJ/mol	Joback Method
hvap	43.44	kJ/mol	Joback Method
log10ws	-3.50		Crippen Method
logp	3.715		Crippen Method
mcvol	161.450	ml/mol	McGowan Method
pc	2248.26	kPa	Joback Method
rinpol	1342.00		NIST Webbook
rinpol	1360.00		NIST Webbook
rinpol	1374.00		NIST Webbook
rinpol	1349.00		NIST Webbook
rinpol	1386.00		NIST Webbook
rinpol	1349.00		NIST Webbook
rinpol	1333.00		NIST Webbook
rinpol	1386.00		NIST Webbook
rinpol	1317.00		NIST Webbook
rinpol	1333.00		NIST Webbook
rinpol	1342.00		NIST Webbook
rinpol	1317.00		NIST Webbook
rinpol	1333.00		NIST Webbook
rinpol	1349.00		NIST Webbook
ripol	1540.00		NIST Webbook
ripol	1582.00		NIST Webbook
ripol	1561.00		NIST Webbook
ripol	1540.00		NIST Webbook
tb	511.55	K	Joback Method
tc	720.55	K	Joback Method
tf	263.09	K	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	417.79	J/molxK	511.55	Joback Method
cpg	521.24	J/molxK	685.72	Joback Method
cpg	503.15	J/molxK	650.88	Joback Method
cpg	483.84	J/molxK	616.05	Joback Method
cpg	463.24	J/molxK	581.22	Joback Method
cpg	441.25	J/molxK	546.38	Joback Method
cpg	538.22	J/molxK	720.55	Joback Method
dvisc	0.0014927	Paxs	511.55	Joback Method
dvisc	0.0014280	Paxs	470.14	Joback Method
dvisc	0.0013545	Paxs	428.73	Joback Method
dvisc	0.0012704	Paxs	387.32	Joback Method
dvisc	0.0011733	Paxs	345.91	Joback Method
dvisc	0.0010605	Paxs	304.50	Joback Method
dvisc	0.0009284	Paxs	263.09	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R134466&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R134466&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>r<sub>inpol</sub>:</b>	Non-polar retention indices
<b>r<sub>ipol</sub>:</b>	Polar retention indices
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature
<b>t<sub>f</sub>:</b>	Normal melting (fusion) point
<b>v<sub>c</sub>:</b>	Critical Volume

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