

# Decane, 1,10-diphenyl-

Inchi:	InChI=1S/C22H30/c1(3-5-9-15-21-17-11-7-12-18-21)2-4-6-10-16-22-19-13-8-14-20-22/h
InchiKey:	SSSBYZTYRWXPNJ-UHFFFAOYSA-N
Formula:	C22H30
SMILES:	c1ccc(CCCCCCCCCc2ccccc2)cc1
Mol. weight [g/mol]:	294.47
CAS:	35511-93-4

## Physical Properties

Property code	Value	Unit	Source
gf	359.18	kJ/mol	Joback Method
hf	-24.35	kJ/mol	Joback Method
hfus	40.82	kJ/mol	Joback Method
hvap	69.12	kJ/mol	Joback Method
log10ws	-7.24		Crippen Method
logp	6.593		Crippen Method
mcvol	273.320	ml/mol	McGowan Method
pc	1404.84	kPa	Joback Method
tb	756.12	K	Joback Method
tc	965.23	K	Joback Method
tf	291.10 ± 1.50	K	NIST Webbook
vc	1.052	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	803.01	J/mol×K	756.12	Joback Method
cpg	822.74	J/mol×K	790.97	Joback Method
cpg	841.21	J/mol×K	825.82	Joback Method
cpg	858.50	J/mol×K	860.68	Joback Method
cpg	874.67	J/mol×K	895.53	Joback Method
cpg	889.82	J/mol×K	930.38	Joback Method
cpg	904.00	J/mol×K	965.23	Joback Method
dvisc	0.0016084	Paxs	390.54	Joback Method
dvisc	0.0006901	Paxs	451.47	Joback Method

dvisc	0.0003621	Paxs	512.40	Joback Method
dvisc	0.0002179	Paxs	573.33	Joback Method
dvisc	0.0001445	Paxs	634.26	Joback Method
dvisc	0.0001031	Paxs	695.19	Joback Method
dvisc	0.0000776	Paxs	756.12	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C35511934&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C35511934&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/28-752-6/Decane-1-10-diphenyl.pdf>

Generated by Cheméo on 2024-04-19 16:23:53.119335215 +0000 UTC m=+15833082.039912530.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.