

# Succinic acid, cis-hex-3-enyl ethyl ester

<b>Inchi:</b>	InChI=1S/C12H20O4/c1-3-5-6-7-10-16-12(14)9-8-11(13)15-4-2/h5-6H,3-4,7-10H2,1-2H3
<b>InchiKey:</b>	WKZZFRSJSZINKQ-WAYWQWQTSA-N
<b>Formula:</b>	C12H20O4
<b>SMILES:</b>	CCC=CCCOC(=O)CCC(=O)OCC
<b>Mol. weight [g/mol]:</b>	228.28

## Physical Properties

Property code	Value	Unit	Source
gf	-337.46	kJ/mol	Joback Method
hf	-663.39	kJ/mol	Joback Method
hfus	32.61	kJ/mol	Joback Method
hvap	60.58	kJ/mol	Joback Method
log10ws	-2.42		Crippen Method
logp	2.229		Crippen Method
mvol	190.520	ml/mol	McGowan Method
pc	2018.13	kPa	Joback Method
rinpol	1564.00		NIST Webbook
rinpol	1564.00		NIST Webbook
tb	630.70	K	Joback Method
tc	814.25	K	Joback Method
tf	364.24	K	Joback Method
vc	0.736	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	495.03	J/molxK	630.70	Joback Method
cpg	509.10	J/molxK	661.29	Joback Method
cpg	522.51	J/molxK	691.88	Joback Method
cpg	535.27	J/molxK	722.48	Joback Method
cpg	547.40	J/molxK	753.07	Joback Method
cpg	558.89	J/molxK	783.66	Joback Method
cpg	569.75	J/molxK	814.25	Joback Method
dvisc	0.0015837	Paxs	364.24	Joback Method

dvisc	0.0008330	Paxs	408.65	Joback Method
dvisc	0.0004970	Paxs	453.06	Joback Method
dvisc	0.0003252	Paxs	497.47	Joback Method
dvisc	0.0002281	Paxs	541.88	Joback Method
dvisc	0.0001688	Paxs	586.29	Joback Method
dvisc	0.0001303	Paxs	630.70	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U353403&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U353403&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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