

# Phenol, 4-butyl-

<b>Other names:</b>	1-(p-Hydroxyphenyl)butane 1-Hydroxy-4-n-butylbenzene 4-Butylphenol 4-n-Butylphenol NSC 407848 Phenol, p-butyl- p-Butylphenol p-Hydroxybutylbenzene p-n-Butylphenol
<b>Inchi:</b>	InChI=1S/C10H14O/c1-2-3-4-9-5-7-10(11)8-6-9/h5-8,11H,2-4H2,1H3
<b>InchiKey:</b>	CYYZDBDROVLTJU-UHFFFAOYSA-N
<b>Formula:</b>	C10H14O
<b>SMILES:</b>	CCCCc1ccc(O)cc1
<b>Mol. weight [g/mol]:</b>	150.22
<b>CAS:</b>	1638-22-8

## Physical Properties

Property code	Value	Unit	Source
gf	-8.89	kJ/mol	Joback Method
hf	-190.51	kJ/mol	Joback Method
hfus	21.48	kJ/mol	Joback Method
hvap	53.14	kJ/mol	Joback Method
ie	8.70 ± 0.10	eV	NIST Webbook
log10ws	-2.66		Crippen Method
logp	2.735		Crippen Method
mcvol	133.870	ml/mol	McGowan Method
pc	3468.36	kPa	Joback Method
rinpol	1386.00		NIST Webbook
rinpol	1360.60		NIST Webbook
rinpol	1364.40		NIST Webbook
rinpol	1335.00		NIST Webbook
rinpol	1360.60		NIST Webbook
rinpol	1335.00		NIST Webbook
rinpol	1361.80		NIST Webbook
ripol	2360.00		NIST Webbook
tb	519.65 ± 3.00	K	NIST Webbook
tb	521.15 ± 3.00	K	NIST Webbook

tb	512.65 ± 3.00	K	NIST Webbook
tb	516.15 ± 4.00	K	NIST Webbook
tb	513.15 ± 3.00	K	NIST Webbook
tb	518.15 ± 5.00	K	NIST Webbook
tb	521.15 ± 3.00	K	NIST Webbook
tb	521.20	K	NIST Webbook
tb	518.15 ± 3.00	K	NIST Webbook
tb	521.15 ± 3.00	K	NIST Webbook
tb	521.15 ± 3.00	K	NIST Webbook
tc	753.87	K	Joback Method
tf	295.15 ± 2.00	K	NIST Webbook
tf	285.15 ± 1.00	K	NIST Webbook
tf	295.15 ± 2.00	K	NIST Webbook
tf	295.15 ± 2.00	K	NIST Webbook
tf	293.65 ± 2.00	K	NIST Webbook
vc	0.454	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	325.39	J/mol×K	571.90	Joback Method
cpg	338.30	J/mol×K	608.29	Joback Method
cpg	350.35	J/mol×K	644.69	Joback Method
cpg	382.17	J/mol×K	753.87	Joback Method
cpg	361.63	J/mol×K	681.08	Joback Method
cpg	372.21	J/mol×K	717.48	Joback Method
cpg	311.56	J/mol×K	535.50	Joback Method
dvisc	0.0000958	Paxs	503.02	Joback Method
dvisc	0.0001614	Paxs	470.53	Joback Method
dvisc	0.0002937	Paxs	438.05	Joback Method
dvisc	0.0005881	Paxs	405.57	Joback Method
dvisc	0.0013291	Paxs	373.08	Joback Method
dvisc	0.0000606	Paxs	535.50	Joback Method
dvisc	0.0035095	Paxs	340.60	Joback Method
hvapt	49.90	kJ/mol	443.00	NIST Webbook
hvapt	54.40	kJ/mol	443.00	NIST Webbook
hvapt	57.60	kJ/mol	443.00	NIST Webbook
hvapt	61.70	kJ/mol	524.00	NIST Webbook
hvapt	56.60	kJ/mol	443.00	NIST Webbook

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	398.70	K	1.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.54562e+01
Coeff. B	-4.71576e+03
Coeff. C	-8.60300e+01
Temperature range (K), min.	396.92
Temperature range (K), max.	550.88

## Sources

McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1638228&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1638228&amp;Units=SI</a>
The Yaws Handbook of Vapor Pressure:	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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