

Bithiopyranylidene, 2,2',6,6'-tetramethyl-

Other names:	4H-Thiopyran, 4-(2,6-dimethyl-4H-thiopyran-4-ylidene)-2,6-dimethyl-4H-Thiine, 2,6-dimethyl-4-(2,6-dimethyl-4H-thiin-4-ylideno)-
Inchi:	InChI=1S/C14H16S2/c1-9-5-13(6-10(2)15-9)14-7-11(3)16-12(4)8-14/h5-8H,1-4H3
InchiKey:	KIVKMHQZJLNIOQ-UHFFFAOYSA-N
Formula:	C14H16S2
SMILES:	<chem>CC1=CC(=C2C=C(C)SC(C)=C2)C=C(C)S1</chem>
Mol. weight [g/mol]:	248.41
CAS:	42506-61-6

Physical Properties

Property code	Value	Unit	Source
gf	303.06	kJ/mol	Joback Method
hf	127.63	kJ/mol	Joback Method
hfus	24.63	kJ/mol	Joback Method
hvap	65.29	kJ/mol	Joback Method
ie	6.23	eV	NIST Webbook
log10ws	-6.49		Crippen Method
logp	5.392		Crippen Method
mcvol	197.600	ml/mol	McGowan Method
pc	2480.12	kPa	Joback Method
tb	689.50	K	Joback Method
tc	951.60	K	Joback Method
tf	516.60	K	Joback Method
vc	0.710	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	483.34	J/molxK	689.50	Joback Method
cpg	499.81	J/molxK	733.18	Joback Method
cpg	515.08	J/molxK	776.87	Joback Method
cpg	529.19	J/molxK	820.55	Joback Method
cpg	542.21	J/molxK	864.23	Joback Method
cpg	554.19	J/molxK	907.91	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C42506616&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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