

# 2H-Pyran-2-methanol, tetrahydro-

<b>Other names:</b>	Tetrahydropyran-2-carbinol Tetrahydropyran-2-methanol Tetrahydropyranyl-2-methanol 2-(Hydroxymethyl)tetrahydropyran Pyran-2-methanol, tetrahydro- 2-Methanol, tetrahydropyran 2-Methylotetrahydro-1,4-pyran 2-Tetrahydropyranilcarbinol Tetrahydro-2H-pyran-2-methanol NSC 5221 tetrahydropyran-2-ylmethanol
<b>Inchi:</b>	InChI=1S/C6H12O2/c7-5-6-3-1-2-4-8-6/h6-7H,1-5H2
<b>InchiKey:</b>	ROTONRWJLXYJBD-UHFFFAOYSA-N
<b>Formula:</b>	C6H12O2
<b>SMILES:</b>	OCC1CCCCO1
<b>Mol. weight [g/mol]:</b>	116.16
<b>CAS:</b>	100-72-1

## Physical Properties

Property code	Value	Unit	Source
gf	-198.85	kJ/mol	Joback Method
hf	-397.08	kJ/mol	Joback Method
hfus	15.20	kJ/mol	Joback Method
hvap	50.57	kJ/mol	Joback Method
log10ws	-0.69		Crippen Method
logp	0.548		Crippen Method
mcvol	96.280	ml/mol	McGowan Method
pc	4409.10	kPa	Joback Method
tb	460.35	K	NIST Webbook
tb	459.95	K	NIST Webbook
tb	460.20	K	NIST Webbook
tc	669.95	K	Joback Method
tf	252.15	K	Joback Method
vc	0.344	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	283.17	J/molxK	669.95	Joback Method
cpg	273.61	J/molxK	637.51	Joback Method
cpg	263.50	J/molxK	605.08	Joback Method
cpg	252.81	J/molxK	572.65	Joback Method
cpg	241.54	J/molxK	540.22	Joback Method
cpg	229.67	J/molxK	507.79	Joback Method
cpg	217.19	J/molxK	475.36	Joback Method
cpl	222.00	J/molxK	298.15	NIST Webbook
dvisc	0.0002250	Paxs	475.36	Joback Method
dvisc	0.0003808	Paxs	438.16	Joback Method
dvisc	0.0007107	Paxs	400.96	Joback Method
dvisc	0.0015067	Paxs	363.75	Joback Method
dvisc	0.0037910	Paxs	326.55	Joback Method
dvisc	0.0120921	Paxs	289.35	Joback Method
dvisc	0.0543129	Paxs	252.15	Joback Method
hvapt	49.00	kJ/mol	402.00	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C100721&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C100721&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>h<sub>vapt</sub>:</b>	Enthalpy of vaporization at a given temperature
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature
<b>t<sub>f</sub>:</b>	Normal melting (fusion) point
<b>v<sub>c</sub>:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/28-809-3/2H-Pyran-2-methanol-tetrahydro.pdf>

Generated by Cheméo on 2024-04-17 02:56:20.352931594 +0000 UTC m=+15611829.273508907.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.