

# Butane, 1-chloro-3-methyl-

<b>Other names:</b>	1-Chloro-3,3-dimethylpropane 1-Chloro-3-methylbutane 3-Methylbutyl chloride 4-Chloro-2-methylbutane Isoamyl chloride Isopentyl chloride
<b>Inchi:</b>	InChI=1S/C5H11Cl/c1-5(2)3-4-6/h5H,3-4H2,1-2H3
<b>InchiKey:</b>	CZHLPWNZCJEPJB-UHFFFAOYSA-N
<b>Formula:</b>	C5H11Cl
<b>SMILES:</b>	CC(C)CCCl
<b>Mol. weight [g/mol]:</b>	106.59
<b>CAS:</b>	107-84-6

## Physical Properties

Property code	Value	Unit	Source
chl	-3355.40 ± 4.20	kJ/mol	NIST Webbook
gf	-23.15	kJ/mol	Joback Method
hf	-171.00 ± 8.80	kJ/mol	NIST Webbook
hfl	-208.00 ± 8.40	kJ/mol	NIST Webbook
hfus	9.38	kJ/mol	Joback Method
hvap	38.10	kJ/mol	NIST Webbook
hvap	36.24	kJ/mol	NIST Webbook
hvap	37.00 ± 1.00	kJ/mol	NIST Webbook
hvap	36.20	kJ/mol	NIST Webbook
hvap	36.33	kJ/mol	NIST Webbook
log10ws	-1.83		Crippen Method
logp	2.271		Crippen Method
mvol	93.550	ml/mol	McGowan Method
pc	3352.86	kPa	Joback Method
rmpol	691.00		NIST Webbook
rmpol	697.00		NIST Webbook
rmpol	723.00		NIST Webbook
rmpol	687.00		NIST Webbook
rmpol	688.00		NIST Webbook
rmpol	706.00		NIST Webbook
rmpol	694.00		NIST Webbook
rmpol	696.00		NIST Webbook

ripol	690.00		NIST Webbook
ripol	880.00		NIST Webbook
ripol	880.00		NIST Webbook
ripol	900.00		NIST Webbook
ripol	886.00		NIST Webbook
tb	372.00	K	NIST Webbook
tb	372.30 ± 0.15	K	NIST Webbook
tb	372.55 ± 0.25	K	NIST Webbook
tb	372.40 ± 2.00	K	NIST Webbook
tb	373.00 ± 0.50	K	NIST Webbook
tb	372.50 ± 0.60	K	NIST Webbook
tb	371.25 ± 1.00	K	NIST Webbook
tc	560.00	K	NIST Webbook
tf	161.03	K	Joback Method
vc	0.358	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	150.99	J/molxK	350.79	Joback Method
cpg	193.96	J/molxK	498.04	Joback Method
cpg	186.05	J/molxK	468.59	Joback Method
cpg	177.80	J/molxK	439.14	Joback Method
cpg	169.22	J/molxK	409.69	Joback Method
cpg	160.28	J/molxK	380.24	Joback Method
cpg	201.54	J/molxK	527.49	Joback Method
dvisc	0.0028647	Paxs	192.66	Joback Method
dvisc	0.0013876	Paxs	224.28	Joback Method
dvisc	0.0008040	Paxs	255.91	Joback Method
dvisc	0.0005253	Paxs	287.54	Joback Method
dvisc	0.0003734	Paxs	319.16	Joback Method
dvisc	0.0078623	Paxs	161.03	Joback Method
dvisc	0.0002823	Paxs	350.79	Joback Method
hvapt	32.30	kJ/mol	368.00	NIST Webbook
hvapt	32.80	kJ/mol	358.00	NIST Webbook
hvapt	33.70	kJ/mol	343.00	NIST Webbook
hvapt	34.60	kJ/mol	328.00	NIST Webbook
hvapt	35.40	kJ/mol	313.00	NIST Webbook
hvapt	32.02	kJ/mol	372.00	NIST Webbook

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.54302e+01
Coeff. B	-3.54120e+03
Coeff. C	-4.44710e+01
Temperature range (K), min.	278.33
Temperature range (K), max.	394.44

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C107846&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C107846&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure

<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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