

# exo-2-Hydroxycineole

<b>Other names:</b>	1,3,3-Trimethyl-2-oxabicyclo[2.2.2]octan-6-ol, (1«alpha»,4«alpha»,6«beta»)-2«alpha»-Hydroxy-1,8-cineole 2-exo-Hydroxy-1,8-cineole 2-Hydroxycineole, exo- trans-2-Hydroxy-1,8-cineole exo-2-Hydroxycineol exo-2-Hydroxy-8-cineole 2«alpha»-Hydroxy-1,8-cineol
<b>Inchi:</b>	InChI=1S/C10H18O2/c1-9(2)7-4-5-10(3,12-9)8(11)6-7/h7-8,11H,4-6H2,1-3H3/t?,8-,10?
<b>InchiKey:</b>	YVCUGZBVCHODNB-ZCUBBSJVSA-N
<b>Formula:</b>	C10H18O2
<b>SMILES:</b>	CC1(C)OC2(C)CCC1CC2O
<b>Mol. weight [g/mol]:</b>	170.25
<b>CAS:</b>	92999-78-5

## Physical Properties

Property code	Value	Unit	Source
gf	-118.72	kJ/mol	Joback Method
hf	-410.88	kJ/mol	Joback Method
hfus	15.34	kJ/mol	Joback Method
hvap	56.29	kJ/mol	Joback Method
log10ws	-2.24		Crippen Method
logp	1.715		Crippen Method
mcvol	141.780	ml/mol	McGowan Method
pc	3224.64	kPa	Joback Method
ripol	1196.00		NIST Webbook
ripol	1219.00		NIST Webbook
ripol	1218.00		NIST Webbook
ripol	1230.00		NIST Webbook
ripol	1228.00		NIST Webbook
ripol	1212.00		NIST Webbook
ripol	1196.00		NIST Webbook
ripol	1228.00		NIST Webbook
ripol	1836.00		NIST Webbook
ripol	1834.00		NIST Webbook
ripol	1840.00		NIST Webbook
ripol	1845.00		NIST Webbook

ripol	1834.00		NIST Webbook
ripol	1875.00		NIST Webbook
ripol	1857.00		NIST Webbook
ripol	1888.00		NIST Webbook
ripol	1861.00		NIST Webbook
ripol	1859.00		NIST Webbook
ripol	1836.00		NIST Webbook
ripol	1870.00		NIST Webbook
tb	560.49	K	Joback Method
tc	767.10	K	Joback Method
tf	358.01	K	Joback Method
vc	0.527	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	386.49	J/mol×K	560.49	Joback Method
cpg	402.41	J/mol×K	594.92	Joback Method
cpg	417.30	J/mol×K	629.36	Joback Method
cpg	431.35	J/mol×K	663.79	Joback Method
cpg	444.77	J/mol×K	698.23	Joback Method
cpg	457.73	J/mol×K	732.66	Joback Method
cpg	470.44	J/mol×K	767.10	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C92999785&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C92999785&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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