

# Thiofanox

<b>Other names:</b>	2-Butanone, 3,3-dimethyl-1-(methylthio)-, O-[(methylamino)carbonyl]oxime 3,3-Dimethyl-1-(Methylthio)-2-butanone O-[(methylamino)carbonyl]oxime 3,3-Dimethyl-1-(methylthio)-2-butanone O-(methylcarbamoyl)oxime 3,3-dimethyl-1-(methylthio)butanone-O-(N-methylcarbamoyl)oxime DS 15647 Dacamox Diamond shamrock DS-15647 ENT 27851 Rcra waste number P045 Thiofanocarb
<b>Inchi:</b>	InChI=1S/C9H18N2O2S/c1-9(2,3)7(6-14-5)11-13-8(12)10-4/h6H2,1-5H3,(H,10,12)
<b>InchiKey:</b>	FZSVSABTBYGOQH-UHFFFAOYSA-N
<b>Formula:</b>	C9H18N2O2S
<b>SMILES:</b>	CNC(=O)ON=C(CSC)C(C)(C)C
<b>Mol. weight [g/mol]:</b>	218.32
<b>CAS:</b>	39196-18-4

## Physical Properties

Property code	Value	Unit	Source
hf	-314.87	kJ/mol	Joback Method
hvap	60.14	kJ/mol	Joback Method
log10ws	-1.62		Aqueous Solubility Prediction Method
log10ws	-1.62		Estimated Solubility Method
logp	2.107		Crippen Method
mcvol	177.120	ml/mol	McGowan Method
pc	2295.91	kPa	Joback Method
rinpol	1274.00		NIST Webbook
rinpol	1279.00		NIST Webbook
tb	673.89	K	Joback Method
tc	895.18	K	Joback Method
tf	331.11 ± 0.20	K	NIST Webbook

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hfust	19.83	kJ/mol	330.20	NIST Webbook
hsubt	94.00 ± 6.00	kJ/mol	313.00	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C39196184&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C39196184&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa</a>
<b>Estimated Solubility Method:</b>	<a href="http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt">http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt</a>

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

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