

# Adipic acid, ethyl pent-4-enyl ester

<b>Inchi:</b>	InChI=1S/C13H22O4/c1-3-5-8-11-17-13(15)10-7-6-9-12(14)16-4-2/h3H,1,4-11H2,2H3
<b>InchiKey:</b>	YXGWLPLHOXJMFK-UHFFFAOYSA-N
<b>Formula:</b>	C13H22O4
<b>SMILES:</b>	C=CCCCOC(=O)CCCC(=O)OCC
<b>Mol. weight [g/mol]:</b>	242.31

## Physical Properties

Property code	Value	Unit	Source
gf	-321.42	kJ/mol	Joback Method
hf	-675.82	kJ/mol	Joback Method
hfus	33.72	kJ/mol	Joback Method
hvap	62.17	kJ/mol	Joback Method
log10ws	-2.84		Crippen Method
logp	2.619		Crippen Method
mcvol	204.610	ml/mol	McGowan Method
pc	1838.84	kPa	Joback Method
rinpol	1659.00		NIST Webbook
rinpol	1659.00		NIST Webbook
tb	646.10	K	Joback Method
tc	824.98	K	Joback Method
tf	378.83	K	Joback Method
vc	0.792	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	545.83	J/molxK	646.10	Joback Method
cpg	560.42	J/molxK	675.91	Joback Method
cpg	574.33	J/molxK	705.73	Joback Method
cpg	587.56	J/molxK	735.54	Joback Method
cpg	600.14	J/molxK	765.35	Joback Method
cpg	612.06	J/molxK	795.17	Joback Method
cpg	623.32	J/molxK	824.98	Joback Method
dvisc	0.0015761	Paxs	378.83	Joback Method

dvisc	0.0008541	Paxs	423.38	Joback Method
dvisc	0.0005201	Paxs	467.92	Joback Method
dvisc	0.0003453	Paxs	512.46	Joback Method
dvisc	0.0002447	Paxs	557.01	Joback Method
dvisc	0.0001825	Paxs	601.55	Joback Method
dvisc	0.0001417	Paxs	646.10	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U353790&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U353790&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/28-977-7/Adipic-acid-ethyl-pent-4-enyl-ester.pdf>

Generated by Cheméo on 2024-04-29 01:53:03.004533172 +0000 UTC m=+16644831.925110488.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.