

# Acetic acid, iso-butenyl ester

<b>Inchi:</b>	InChI=1S/C6H10O2/c1-5(2)4-8-6(3)7/h4H,1-3H3
<b>InchiKey:</b>	MNJRLZXXSZEIHD-UHFFFAOYSA-N
<b>Formula:</b>	C6H10O2
<b>SMILES:</b>	CC(=O)OC=C(C)C
<b>Mol. weight [g/mol]:</b>	114.14
<b>CAS:</b>	14478-14-9

## Physical Properties

Property code	Value	Unit	Source
gf	-162.61	kJ/mol	Joback Method
hf	-304.54	kJ/mol	Joback Method
hfus	12.97	kJ/mol	Joback Method
hvap	38.14	kJ/mol	Joback Method
log10ws	-1.55		Crippen Method
logp	1.473		Crippen Method
mcvol	98.540	ml/mol	McGowan Method
pc	3497.14	kPa	Joback Method
tb	417.01	K	Joback Method
tc	607.33	K	Joback Method
tf	210.50	K	Joback Method
vc	0.377	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	184.26	J/molxK	417.01	Joback Method
cpg	193.90	J/molxK	448.73	Joback Method
cpg	203.14	J/molxK	480.45	Joback Method
cpg	212.00	J/molxK	512.17	Joback Method
cpg	220.47	J/molxK	543.89	Joback Method
cpg	228.57	J/molxK	575.61	Joback Method
cpg	236.31	J/molxK	607.33	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C14478149&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C14478149&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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