

# Succinic acid, ethyl 2,3,6-trichlorophenyl ester

**Inchi:** InChI=1S/C12H11Cl3O4/c1-2-18-9(16)5-6-10(17)19-12-8(14)4-3-7(13)11(12)15/h3-4H,2  
**InchiKey:** VQAWDMHUBWVGEG-UHFFFAOYSA-N  
**Formula:** C12H11Cl3O4  
**SMILES:** CCOC(=O)CCC(=O)Oc1c(Cl)ccc(Cl)c1Cl  
**Mol. weight [g/mol]:** 325.57

## Physical Properties

Property code	Value	Unit	Source
gf	-369.95	kJ/mol	Joback Method
hf	-625.71	kJ/mol	Joback Method
hfus	37.87	kJ/mol	Joback Method
hvap	78.03	kJ/mol	Joback Method
log10ws	-4.38		Crippen Method
logp	3.896		Crippen Method
mcvol	207.780	ml/mol	McGowan Method
pc	2261.11	kPa	Joback Method
rinpol	2100.00		NIST Webbook
rinpol	2100.00		NIST Webbook
tb	780.45	K	Joback Method
tc	1003.15	K	Joback Method
tf	523.06	K	Joback Method
vc	0.794	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	503.14	J/molxK	780.45	Joback Method
cpg	513.26	J/molxK	817.57	Joback Method
cpg	522.53	J/molxK	854.68	Joback Method
cpg	530.96	J/molxK	891.80	Joback Method
cpg	538.53	J/molxK	928.92	Joback Method
cpg	545.24	J/molxK	966.04	Joback Method
cpg	551.09	J/molxK	1003.15	Joback Method
dvisc	0.0005598	Paxs	523.06	Joback Method

dvisc	0.0003857	Paxs	565.96	Joback Method
dvisc	0.0002801	Paxs	608.86	Joback Method
dvisc	0.0002121	Paxs	651.75	Joback Method
dvisc	0.0001663	Paxs	694.65	Joback Method
dvisc	0.0001341	Paxs	737.55	Joback Method
dvisc	0.0001107	Paxs	780.45	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U349701&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U349701&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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