

# Adipic acid, 2-chloropropyl octyl ester

<b>Inchi:</b>	InChI=1S/C17H31ClO4/c1-3-4-5-6-7-10-13-21-16(19)11-8-9-12-17(20)22-14-15(2)18/h15
<b>InchiKey:</b>	XRQJJAULZVTGRQ-UHFFFAOYSA-N
<b>Formula:</b>	C17H31ClO4
<b>SMILES:</b>	CCCCCCCCOC(=O)CCCCC(=O)OCC(C)Cl
<b>Mol. weight [g/mol]:</b>	334.88

## Physical Properties

Property code	Value	Unit	Source
gf	-389.95	kJ/mol	Joback Method
hf	-904.83	kJ/mol	Joback Method
hfus	46.03	kJ/mol	Joback Method
hvap	75.75	kJ/mol	Joback Method
log10ws	-4.93		Crippen Method
logp	4.621		Crippen Method
mvol	277.510	ml/mol	McGowan Method
pc	1287.44	kPa	Joback Method
rmpol	2241.00		NIST Webbook
tb	777.93	K	Joback Method
tc	962.16	K	Joback Method
tf	440.59	K	Joback Method
vc	1.079	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	822.99	J/molxK	777.93	Joback Method
cpg	839.25	J/molxK	808.63	Joback Method
cpg	854.58	J/molxK	839.34	Joback Method
cpg	869.00	J/molxK	870.04	Joback Method
cpg	882.53	J/molxK	900.75	Joback Method
cpg	895.16	J/molxK	931.45	Joback Method
cpg	906.92	J/molxK	962.16	Joback Method
dvisc	0.0010819	Paxs	440.59	Joback Method
dvisc	0.0005194	Paxs	496.81	Joback Method

dvisc	0.0002895	Paxs	553.04	Joback Method
dvisc	0.0001797	Paxs	609.26	Joback Method
dvisc	0.0001209	Paxs	665.48	Joback Method
dvisc	0.0000866	Paxs	721.71	Joback Method
dvisc	0.0000650	Paxs	777.93	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U353544&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U353544&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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