

Glutaric acid, 2,6-dichlorophenyl isoheptyl ester

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|----------------------|---|
| Inchi: | InChI=1S/C17H22Cl2O4/c1-12(2)6-5-11-22-15(20)9-4-10-16(21)23-17-13(18)7-3-8-14(1) |
| InchiKey: | OTHBCYABNYWBIR-UHFFFAOYSA-N |
| Formula: | C17H22Cl2O4 |
| SMILES: | CC(C)CCCOC(=O)CCCC(=O)Oc1c(Cl)cccc1Cl |
| Mol. weight [g/mol]: | 361.26 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -308.73 | kJ/mol | Joback Method |
| hf | -706.98 | kJ/mol | Joback Method |
| hfus | 43.49 | kJ/mol | Joback Method |
| hvap | 83.73 | kJ/mol | Joback Method |
| log10ws | -5.54 | | Crippen Method |
| logp | 5.048 | | Crippen Method |
| mcvol | 265.990 | ml/mol | McGowan Method |
| pc | 1570.96 | kPa | Joback Method |
| rinpol | 2484.00 | | NIST Webbook |
| rinpol | 2484.00 | | NIST Webbook |
| tb | 852.00 | K | Joback Method |
| tc | 1063.90 | K | Joback Method |
| tf | 521.97 | K | Joback Method |
| vc | 1.020 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 755.82 | J/molxK | 852.00 | Joback Method |
| cpg | 768.99 | J/molxK | 887.32 | Joback Method |
| cpg | 781.08 | J/molxK | 922.63 | Joback Method |
| cpg | 792.11 | J/molxK | 957.95 | Joback Method |
| cpg | 802.10 | J/molxK | 993.26 | Joback Method |
| cpg | 811.05 | J/molxK | 1028.58 | Joback Method |
| cpg | 818.99 | J/molxK | 1063.90 | Joback Method |
| dvisc | 0.0005217 | Paxs | 521.97 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0003039 | Paxs | 576.98 | Joback Method |
| dvisc | 0.0001945 | Paxs | 631.98 | Joback Method |
| dvisc | 0.0001337 | Paxs | 686.99 | Joback Method |
| dvisc | 0.0000971 | Paxs | 741.99 | Joback Method |
| dvisc | 0.0000738 | Paxs | 797.00 | Joback Method |
| dvisc | 0.0000580 | Paxs | 852.00 | Joback Method |

Sources

| | |
|------------------------|---|
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U358832&Units=SI |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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