

# Diethylmalonic acid, octyl 2,4,4-trimethylpentyl ester

Inchi:	InChI=1S/C23H44O4/c1-8-11-12-13-14-15-16-26-20(24)23(9-2,10-3)21(25)27-18-19(4)1
InchiKey:	SGJWOVFB DJTFSN-UHFFFAOYSA-N
Formula:	C23H44O4
SMILES:	CCCCCCCCOC(=O)C(CC)(CC)C(=O)OCC(C)CC(C)(C)C
Mol. weight [g/mol]:	384.59

## Physical Properties

Property code	Value	Unit	Source
gf	-321.82	kJ/mol	Joback Method
hf	-1030.43	kJ/mol	Joback Method
hfus	42.55	kJ/mol	Joback Method
hvap	82.12	kJ/mol	Joback Method
log10ws	-6.45		Crippen Method
logp	6.312		Crippen Method
mcvol	349.810	ml/mol	McGowan Method
pc	924.99	kPa	Joback Method
rinpola	2210.00		NIST Webbook
rinpola	2210.00		NIST Webbook
tb	871.32	K	Joback Method
tc	1068.48	K	Joback Method
tf	483.13	K	Joback Method
vc	1.343	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1155.56	J/molxK	871.32	Joback Method
cpg	1240.53	J/molxK	1035.62	Joback Method
cpg	1225.72	J/molxK	1002.76	Joback Method
cpg	1209.88	J/molxK	969.90	Joback Method
cpg	1192.94	J/molxK	937.04	Joback Method
cpg	1174.86	J/molxK	904.18	Joback Method
cpg	1254.37	J/molxK	1068.48	Joback Method
dvisc	0.0000192	Paxs	871.32	Joback Method

dvisc	0.0000271	Paxs	806.62	Joback Method
dvisc	0.0000405	Paxs	741.92	Joback Method
dvisc	0.0000656	Paxs	677.22	Joback Method
dvisc	0.0001173	Paxs	612.53	Joback Method
dvisc	0.0002409	Paxs	547.83	Joback Method
dvisc	0.0005997	Paxs	483.13	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369480&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369480&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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