

# Propane, 2,2'-[ethylidenebis(oxy)]bis-

<b>Other names:</b>	1,1-Diisopropoxyethane 2,4,6-Trimethyl-3,5-dioxaheptane 2-(1-propan-2-yloxyethoxy)propane
<b>Inchi:</b>	InChI=1S/C8H18O2/c1-6(2)9-8(5)10-7(3)4/h6-8H,1-5H3
<b>InchiKey:</b>	TWWSMHPNERSWRN-UHFFFAOYSA-N
<b>Formula:</b>	C8H18O2
<b>SMILES:</b>	CC(C)OC(C)OC(C)C
<b>Mol. weight [g/mol]:</b>	146.23
<b>CAS:</b>	4285-59-0

## Physical Properties

Property code	Value	Unit	Source
chl	-5151.30 ± 3.50	kJ/mol	NIST Webbook
gf	-200.84	kJ/mol	Joback Method
hf	-525.90 ± 4.80	kJ/mol	NIST Webbook
hfl	-569.20 ± 3.50	kJ/mol	NIST Webbook
hfus	8.28	kJ/mol	Joback Method
hvap	43.30	kJ/mol	NIST Webbook
log10ws	-2.18		Crippen Method
logp	2.182		Crippen Method
mcvol	135.320	ml/mol	McGowan Method
pc	2510.03	kPa	Joback Method
tb	395.00 ± 3.00	K	NIST Webbook
tc	601.27	K	Joback Method
tf	179.38	K	Joback Method
vc	0.501	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	282.46	J/mol×K	425.96	Joback Method
cpg	296.02	J/mol×K	455.18	Joback Method
cpg	309.17	J/mol×K	484.40	Joback Method
cpg	321.92	J/mol×K	513.62	Joback Method

cpg	334.26	J/mol×K	542.83	Joback Method
cpg	346.18	J/mol×K	572.05	Joback Method
cpg	357.69	J/mol×K	601.27	Joback Method
dvisc	0.0165879	Paxs	179.38	Joback Method
dvisc	0.0037875	Paxs	220.48	Joback Method
dvisc	0.0013755	Paxs	261.57	Joback Method
dvisc	0.0006577	Paxs	302.67	Joback Method
dvisc	0.0003752	Paxs	343.77	Joback Method
dvisc	0.0002413	Paxs	384.86	Joback Method
dvisc	0.0001689	Paxs	425.96	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4285590&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4285590&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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