

# trans-Perfluorodecalin

<b>Inchi:</b>	InChI=1S/C10F18/c11-1-2(12,5(17,18)9(25,26)7(21,22)3(1,13)14)6(19,20)10(27,28)8(23
<b>InchiKey:</b>	UWEYRJFJVCLAGH-IJWZVTFUSA-N
<b>Formula:</b>	C10F18
<b>SMILES:</b>	FC1(F)C(F)(F)C(F)(F)C2(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F)C2(F)C1(F)F
<b>Mol. weight [g/mol]:</b>	462.08
<b>CAS:</b>	60433-12-7

## Physical Properties

Property code	Value	Unit	Source
chl	-3466.70 ± 3.90	kJ/mol	NIST Webbook
gf	-3516.74	kJ/mol	Joback Method
hf	-3644.50 ± 3.90	kJ/mol	NIST Webbook
hfl	-3689.90 ± 3.90	kJ/mol	NIST Webbook
hfus	10.55	kJ/mol	Joback Method
hvap	45.90 ± 0.60	kJ/mol	NIST Webbook
hvap	45.40 ± 0.10	kJ/mol	NIST Webbook
hvap	45.40 ± 0.08	kJ/mol	NIST Webbook
hvap	45.40 ± 0.10	kJ/mol	NIST Webbook
log10ws	-6.24		Crippen Method
logp	5.513		Crippen Method
mvol	161.900	ml/mol	McGowan Method
pc	1676.91	kPa	Joback Method
sl	517.10	J/molxK	NIST Webbook
tb	410.66	K	Joback Method
tc	541.15	K	Joback Method
tf	439.96	K	Joback Method
tt	294.61 ± 0.01	K	NIST Webbook
vc	0.773	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	465.83	J/molxK	454.16	Joback Method
cpg	478.31	J/molxK	475.90	Joback Method

cpg	489.27	J/mol×K	497.65	Joback Method
cpg	498.89	J/mol×K	519.40	Joback Method
cpg	435.55	J/mol×K	410.66	Joback Method
cpg	451.64	J/mol×K	432.41	Joback Method
cpg	507.37	J/mol×K	541.15	Joback Method
cpl	446.80	J/mol×K	298.15	NIST Webbook
hfust	17.96	kJ/mol	294.61	NIST Webbook
hfust	17.96	kJ/mol	294.60	NIST Webbook
hvapt	43.30	kJ/mol	366.00	NIST Webbook
sfust	61.09	J/mol×K	294.61	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.37219e+01
Coeff. B	-3.08882e+03
Coeff. C	-7.53520e+01
Temperature range (K), min.	305.27
Temperature range (K), max.	442.61

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C60433127&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C60433127&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity

<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>sl:</b>	Liquid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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