

# Benzoic acid, 2-methyl-, anhydride

<b>Other names:</b>	2-Methylbenzoic acid anhydride
<b>Inchi:</b>	InChI=1S/C16H14O3/c1-11-7-3-5-9-13(11)15(17)19-16(18)14-10-6-4-8-12(14)2/h3-10H,1
<b>InchiKey:</b>	YLBSXJWDERHYFY-UHFFFAOYSA-N
<b>Formula:</b>	C16H14O3
<b>SMILES:</b>	<chem>Cc1cccc1C(=O)OC(=O)c1cccc1C</chem>
<b>Mol. weight [g/mol]:</b>	254.28
<b>CAS:</b>	607-86-3

## Physical Properties

Property code	Value	Unit	Source
chs	-7763.40 ± 7.90	kJ/mol	NIST Webbook
gf	-73.44	kJ/mol	Joback Method
hf	-280.83	kJ/mol	Joback Method
hfs	-533.50 ± 7.90	kJ/mol	NIST Webbook
hfus	28.89	kJ/mol	Joback Method
hvap	72.99	kJ/mol	Joback Method
log10ws	-4.63		Crippen Method
logp	3.301		Crippen Method
mcvol	197.790	ml/mol	McGowan Method
pc	2467.81	kPa	Joback Method
tb	758.96	K	Joback Method
tc	998.86	K	Joback Method
tf	470.05	K	Joback Method
vc	0.746	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	531.45	J/molxK	758.96	Joback Method
cpg	545.35	J/molxK	798.94	Joback Method
cpg	558.06	J/molxK	838.93	Joback Method
cpg	569.64	J/molxK	878.91	Joback Method
cpg	580.13	J/molxK	918.89	Joback Method
cpg	589.55	J/molxK	958.87	Joback Method

cpg	597.96	J/molxK	998.86	Joback Method
dvisc	0.0008599	Paxs	470.05	Joback Method
dvisc	0.0005355	Paxs	518.20	Joback Method
dvisc	0.0003615	Paxs	566.35	Joback Method
dvisc	0.0002595	Paxs	614.50	Joback Method
dvisc	0.0001955	Paxs	662.66	Joback Method
dvisc	0.0001530	Paxs	710.81	Joback Method
dvisc	0.0001235	Paxs	758.96	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C607863&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C607863&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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